Enzymes

Return please

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Deoxyribose   DNA   (Thymine)

Ribose   RNA   Ad, Guanine, Uracil, Cibon

(ABC)
(ABB)

[RNA]

(TNR ⊕) $p =$ (10⁻⁴)   (10⁻⁴)

$\underline{RNA+}$

PO—AA

P—O—AA

$3 \cdot 10^3$ cm/sec $1000 = m$

100

1000

$= \left(\frac{\frac{3}{2} \cdot 0 - 1 \cdot \frac{1}{4} \cdot \sqrt{2} \cdot 4}{\frac{3}{2} \cdot 0 H} \right)^{1/2} = \left(\frac{\sqrt{2}/2 - 0}{\sqrt{2}/2} \right)^{1/2}$

$(1 - 0 - 1)$

Enzymes

1457

Suppose now ######## T ϕ
$E_R^{\#}$ mutates back to E_R ; Now there
is not enough repressor $(T_r - R)$ produced
And balance is upset . But now
another foreign Template M_2 can mutate
to T_2 producing E_2

$$(T_r(1) - R) \quad E_2 \quad (T_r(2) - R)$$

$$\boxed{T - E_2 - (T_r(2) - R)}$$

controls self
and contracts E_1
represses

$$\boxed{T - E_1 - (T_r(2) - R)}$$

If now T mutates ### again to perform
yield $E_0^{\#}$ instead of E

optionally arise again a heresy (gene
may arise producing an E_3
which deta leads from $(T_r(2) - R)$ to
$(T_r(3) - R)$ which represses production
of E_2 . etc. etc. ad infinitum

Prior to mutation

2nd + Serine $\xrightarrow{E (Tr)}$ Tryptophane $Tr + R \xrightarrow{E_R} (Tr - R)$
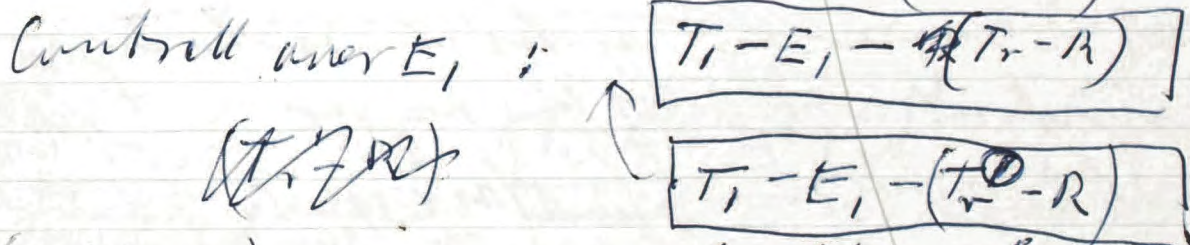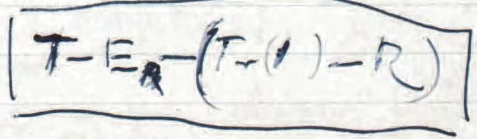
$$T - E - (Tr - R)$$

$$T - E_R - (Tr - R)$$

$E(Tr)$

Mutation occurs in the template of E_R as a result in place of enzyme E_R we have E_R^* and weak binding

this leads to production of

$$T - E_R^* \sim (Tr - R) \quad \text{weak}$$

too much repressor $(Tr - R)$. As a result of too much $(Tr - R)$ enzyme E is repressed.

Now suppressor mutation occurs in a foreign gene, the new enzyme Produced is called E_1, catalysis reaction;

$$(Tr - R) \xrightarrow{E_1} (Tr^{①} - R)$$

Control over E_1 ;

$$T_1 - E_1 - (Tr - R)$$

$$T_1 - E_1 - (Tr^{①} - R)$$

$(Tr^{①} - R)$ represses production of E

$$T - E_R - (Tr - ① - R)$$

write AA (1)
AA (2)
note:
E_0 for E_R
write
for $(Tr(①) - R$
Rep①

overall

and also

$$\boxed{V - E(0) - Rep(2)}$$

$$\boxed{T - E(1) - Rep(2)}$$

AA level may now be restored to normal but now up again $E(0)$ mutates to $E^*(0)$. Now again too little AA is made but level can be restored by a third suppressor mutation in which an enzyme E_3 makes its appearance

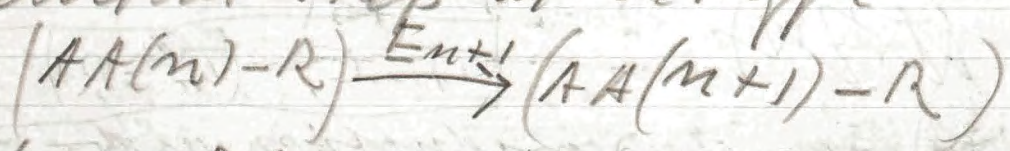$$(AA(2) - R) \xrightarrow{E_3} (AA(3) - R) \quad repr(3)$$

overall

$$\boxed{T - E(3) - repr(3)}$$

$$\boxed{T - E(2) - repr(3)}$$

Now AA level may be just right until E^* again a mutation in $T(0)$ again changes $E^*(0)$ into $E(0)$

Thus there will be an accumulation of Enzymes which catalyze a biochemical step of the type
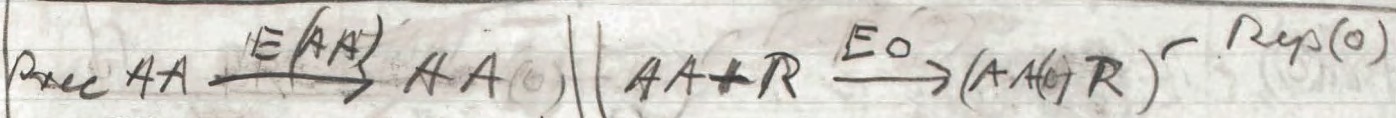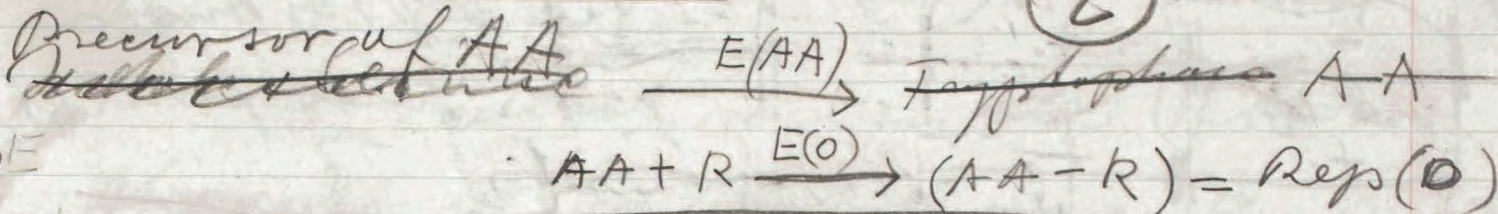
$$(AA(n) - R) \xrightarrow{E_{n+1}} (AA(n+1) - R)$$

Exploring Shuster

Remark first about

adapt. Enz formation

Class I enzymes being directly involved in regulating level of enzymes

pathways involved

class II

$$\text{Precursor of } AA \xrightarrow{E(AA)} \text{Tryptophan } AA$$

$$AA + R \xrightarrow{E(0)} (AA - R) = Rep(0)$$

$$\text{Prec } AA \xrightarrow{E(AA)} AA(0) \,\| \, AA + R \xrightarrow{E0} (AA(0) R)^{\frown Rep(0)}$$

$\boxed{T - E(AA) - Rep(0)}$ $\boxed{T - E_0 - Rep(0)}$
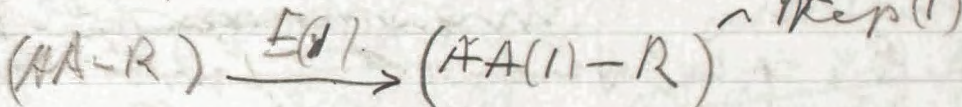
Template making E_0 mutates and makes E_0^* bind weakly to $Rep \, 0$ in the (then E_0^*) sense that   high diss. rate

$\boxed{T - E_0^* \sim Rep(0)}$ (to blike $E(AA)$ will be made and we have a slow grower or auxotroph requiring AA as a growth build better

therefore to much $Rep(0)$ will be made
~~this is now remedied in a suppressor mutation~~

A Foreign template ~~making~~ now mutates and make
~~protein now an~~ an enzyme E_1 which does this

$$(AA - R) \xrightarrow{E(1)} (AA(1) - R)^{\frown Rep(1)}$$

Where $AA(1)$ is one step removed from the original AA.   (theorem)

control $\boxed{T - E(1) - Rep(1)}$

but also   strong binding

$\boxed{T - E^*(0) - Rep(1)}$

Therefore now ~~since~~ $E^*(0)$ being repressed
less ~~too little~~ $Rep(0)$ is made ~~and too much~~
~~AA is made~~ and everything is fine, until E_0^* mut E_0
~~Then now too much~~ AA is made
~~this is~~ remedied by a second suppressor
mutation in another gene
same template mutates   to produce an enzym E_2

$$(AA(1) - R) \xrightarrow{E_2} (AA(2) - R)^{\frown step \, ②}$$

(5) gene cooperation (nuclear gene [?]) (shifts after time...)

$E(AA(0))$ $E(AA(1))$ E_0 ... rep(0)

$$AA(0) + R \xrightarrow{E_0} (AA(0) - R)$$

$$AA(0) \xrightarrow{E_{AA(1)}} \boxed{AA(1)}$$

Differentiation at end //

Is in this general system a situation where a substrate is not produced (except perhaps at a very low level) but if once level is raised more stable equilibration is reached. Is it stable or do fluctuations throw it back [?] to low level. Is this sort of arguing?

When only few enzymes are

take a substrate

S_n which resembles (AA n

there is in the huset. ~~and~~ see E_{n+1} rep(n+1)
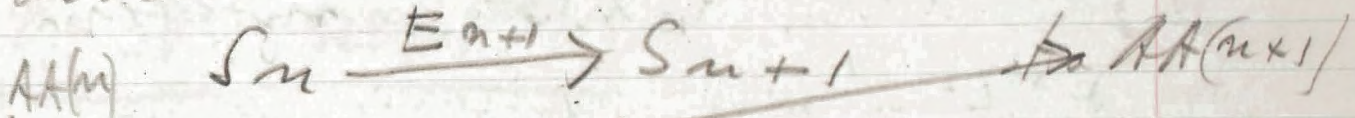
$$(AA (n) - R) \xrightarrow{E_{n+1}} (AA (n+1) - R)$$

$$\boxed{T - E(n+1) - rep(n+1)}$$

$$\boxed{T - E(n+1) - rep(n)}$$

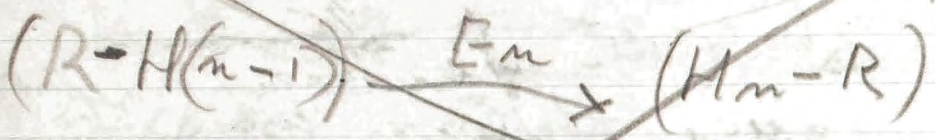$$\boxed{T - E(n+1) - rep(n+2)}$$

~~rep(n+1)~~

We assume now

$$AA(n) \quad S_n \xrightarrow{E_{n+1}} S_{n+1} \quad \xcancel{\longrightarrow AA(n+1)}$$

AA/ Because of the chemical analogy
of S_n or S_{n+1}, ~~thus~~ individual the
enzymes $E(n-1)$; $E(n+1)$ or $E(n+2)$
which make rep(n), rep(n+1) or rep(n+2)
and thus we may expect
that the enzyme E_{n+1} will ride
when the huset grow in the presence
of S_n which resembles AA(n)

① Antibodies

$(R - H) \, E_n \, (H_n - R)$
   $n-1$

$(R \to H(n-1)) \xrightarrow{E_n} (H_n - R)$ $\swarrow rep(n)$

irreversible $E_n \to (H\text{-prot}) \, antigen$

requilibr. $H \to A \, N \, (H_n - R)$

$A \, N \, H_n$
$T \, N \, R$

cause of rep(n)
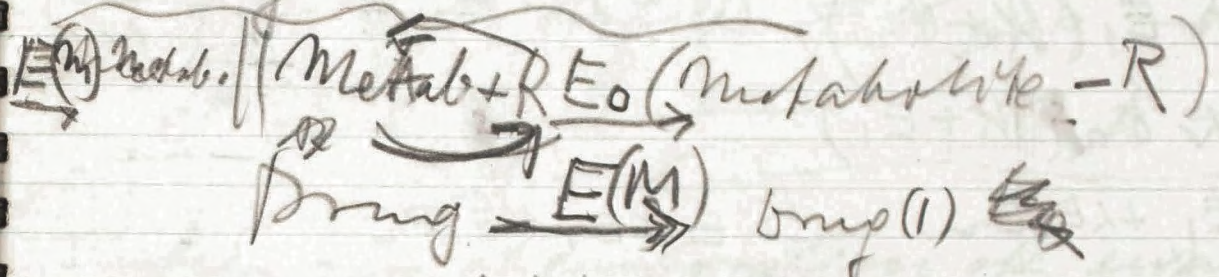
$\uparrow \, L \, A \, \downarrow (rep(n))$

more Metab-

made since evapo- Ⓗ
ration is a <s>st</s> first order
process sometimes more
is made and will falls
back in low equilibrium.

Drug tolerance: ————— repressor (M)

$\boxed{E}$ (M) metab. || Metab + R E_0 (Metabolite - R)

Drug $\xrightarrow{E(M)}$ drug (1)

of Drug which is chemical analog of M
inhibits E_0 which produces repressor.

$$\boxed{T - E(M) - \text{Rep}(M)}$$

Symbols

$$\begin{array}{c} E - A(O) \\ | \quad | \\ T - R \end{array} \text{ or } \begin{array}{c} E \\ | \\ T \end{array} \rangle \text{ rep.}_{AA} \begin{array}{c} E - Hm \\ | \quad | \\ T - R \end{array}$$

Let us set $\qquad \alpha = \beta \qquad a = b \qquad A = B$

$$\begin{cases} 0 = \dfrac{k}{k + E_2} - \lambda E_1 \\[4mm] 0 = \dfrac{k}{k + E_1} - \lambda E_2 \end{cases} \qquad \begin{array}{l} \alpha E_2 = E_2 \\[1mm] \beta E_1 = E_1 \\[1mm] \dfrac{A}{a} = \lambda \\[1mm] k_1 = k_2 = k \end{array}$$

(maximum value possible for $E_1 = \dfrac{a}{A} \left(= \dfrac{1}{\lambda} \right)$

for $E_2 = \dfrac{b}{B} \left(= \dfrac{1}{\lambda} \right)$

$$\begin{cases} k = \lambda E_1 (k + E_2) \\[2mm] k = \lambda E_2 (k + E_1) \\ \hline \end{cases}$$

$$\left. \begin{array}{ll} k = \lambda k E_1 + \lambda E_1 E_2 & \dfrac{k}{\lambda} = k E_1 + E_1 E_2 \\[4mm] k = \lambda k E_2 + \lambda E_1 E_2 & \dfrac{k}{\lambda} = k E_2 + E_1 E_2 \end{array} \right\}$$

clearly not possible ~~AA~~ except
for $E_1 = E_2$ for
if we subtract the two
from each other
$$\dfrac{k}{\lambda}$$
$$0 = k(E_2 - E_1)$$
or $E_2 = E_1$

Now what is rate of dehidro-
ration ?
All this assumes no control
by own substrate so far

Differentiation

Enzyme Kinetics

S substrate

e tot. enzyme

e* free enzyme $= e - p$

p complex [ES]

equilibrium

rate of dissoc. of [ES]

$$K_{1} e^{*} s = K_{2} p$$

$$e^{*} = e \frac{k}{k + s} \qquad k = \frac{K_{2}}{K_{1}}$$

when rate of evaporation of enzyme is high (weak binding)   k is large

Differentiation:

T_1 makes E_1 at rate aX

conc of free template

$$\frac{dE_1}{dt} = a \frac{k_1}{k_1 + \alpha E_2} - A E_1$$

αE_2 is conc. of product of E_2

$$\frac{dE_2}{dt} = b \frac{k_2}{k_2 + \beta E_1} - B E_2$$

Is there a stationary state

$$0 = a \frac{k_1}{k_1 + \alpha E_2} - A E_1$$

$$0 = b \frac{k_2}{k_2 + \alpha E_1} - B E_2$$

$$E/X = V$$

$$L = \frac{1}{10} + \frac{k}{k}\frac{1}{10}E_2 + \frac{k}{\ell}\frac{1}{100}$$

$$\frac{1}{10} = X$$

$$V = \frac{1}{X} + \frac{1}{X} - \frac{1}{XX} + \frac{k}{\ell}\frac{1}{X^2}$$

$$E_2 = \frac{1}{?} - \left[\frac{k+1}{R}\frac{1}{X} + \frac{1}{\ell}\frac{1}{X^2}\right.$$

$$1 = \frac{k}{X} + \frac{1}{X}\frac{k}{R}$$

$$\left\{\begin{array}{l} \dfrac{k}{\lambda} = kE_1 + E_1 E_2 + \dfrac{k}{\ell}E_1^2 \\[2em] \dfrac{k}{\lambda} = kE_2 + E_1 E_2 + \dfrac{k}{\ell}E_2^2 \end{array}\right. \qquad \dfrac{k}{\lambda}\dfrac{1}{E_1} = R + E_2 + \dfrac{k}{\ell}E_1$$

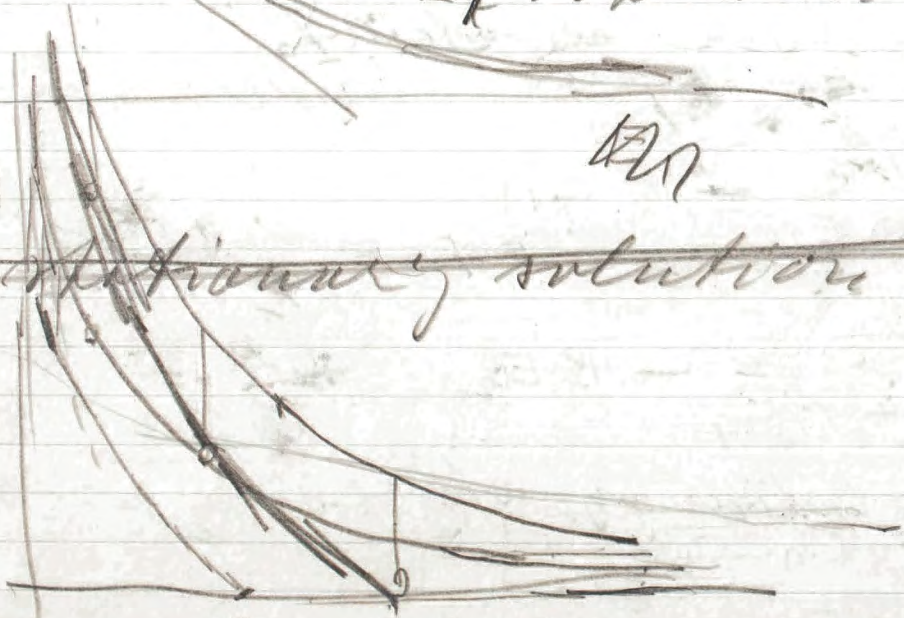$$E_2 = \frac{1}{E_1}\frac{k}{\lambda} - \left[k + \frac{k}{\ell}E_1\right] \qquad E_2 = \frac{1}{E_1}\frac{k}{\lambda}$$

$$E_1 = \frac{1}{E_2}\frac{k}{\lambda} - \left[k + \frac{k}{\ell}E_2\right]$$

Now express from here
E_1 as function of E_2

No stationary solution!

No stationary solution
it seems.

Differentiation (double interaction
stationary?
(Control by own substrate also)

$$\begin{cases} k_1 * e^*{}_{\times} s_1 = k_3 p_1 \\ k_2 {}_{\times} e^* {}_{\times} s_2 = k_3 p_2 \end{cases} \qquad p_1 + p_2 + e^* = e$$

guess $e^* = e \dfrac{1}{1 + \frac{s_1}{k_1} + \frac{s_2}{k_2}}$ check! $k_1 = \dfrac{k_3}{k_1}$

$\dfrac{1}{k_1} = g_1$

$\dfrac{1}{k_2} = g_2$

$$\begin{cases} e^* s_1 = k_1 p_1 \\ e^* s_2 = k_2 p_2 \end{cases}$$

Condition for stationary state now
become if ~~binding~~ constant for
own Template is not k but ℓ

$$\begin{cases} 0 = \dfrac{1}{1 + \frac{E_2}{k} + \frac{E_1}{\ell}} - \lambda E_1 \\[4mm] 0 = \dfrac{1}{1 + \frac{E_1}{k} + \frac{E_1}{\ell}} - \kappa E_2 \end{cases}$$

$\boxed{\ell \gg k}$

$$\begin{cases} 1 = \lambda E_1 + \frac{\lambda}{k} E_1 E_2 + \frac{\lambda}{\ell} E_1^2 \\[2mm] 1 = \lambda E_2 + \frac{\lambda}{k} E_1 E_2 + \frac{\lambda}{\ell} E_2^2 \end{cases}$$

$$1 = \lambda (E_2 - E_1) + \frac{\lambda}{\ell}(E_2 - E_1)(E_2 + E_1)$$

RNA turnover in Brain
Hyden

Antibodies (Chart)

(1)

$$[R-H_m] \xrightarrow[E_m]{E_m} (H_{m+1}-R)^{rep(H)}$$

$$\left| \begin{array}{l} antigen = \\ (Prot-H) \end{array} \right.$$

$$Rev: \quad A \sim H(n)$$

$$E \sim (H_{m+1} R); \quad \frac{1}{T} \sim R$$

Inv: $(Prot-H) + E = (Prot-H) = E$ dead

If it gets to lt. $\begin{array}{c} E-H \\ T-Prot \end{array}$ If it gets to it $\begin{array}{c} A-H \\ T-Prot \end{array}$

(2)

before

3

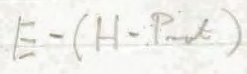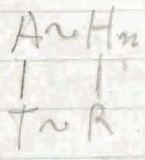after 6 days   insert

put in two

compost
men

rate of dying:

Test of theory
x ray distribution of
the two enzyme in
proportion of amount present
what does this yield?

Gaussian distribution of
R might give function

function $+g$

$$E_1 \longrightarrow R \cdot g_1$$

Presentation $R - g$ $\quad E_2 \quad R - g_2$

immediately after antigen is added

②

$E_m (H - \text{Part})$

$E_m \quad E_m$
$\dfrac{}{+} \quad \dfrac{}{+}$

$\dfrac{A \sim H_m}{1} \Big| \dfrac{A \sim H_m}{1}$
$\dfrac{1}{T \sim R} \Big| \dfrac{1}{T \sim R}$

$\dfrac{A \sim H_n}{\dfrac{1}{T \sim R}}$ $\qquad E - (H - \text{Part})$

conc/ rep $(H\bar{n}\,R)$ $= r$ e concentration of

$$\frac{dr}{dt} = A e - \frac{r}{\tau(r)}$$

$$\frac{dr}{dt} = \frac{1}{\alpha}\,{}^E$$

$$\frac{de}{dt} = B - \frac{e}{\tau(E)}$$

$$\frac{1}{B} = \tau(E) \sim month$$

$$B = e_1$$

$$e = e_1 (1 - e^{-\beta t})$$

$$\frac{dr}{dt} = \underbrace{A e_1}_{const} (1 - e^{-\beta t}) - \alpha r$$

Rate of Antibody
prod. is prop. to free
Template. T

$$[T] = \frac{k(r)}{k(r) + r}$$

Single Interactions, again '2

After several week

Explain Anamnestic response

Tolerance

in embryo or young Rabbit
if antigen is given

A — H
| |
T — Prot

Gal + ATP $\xrightarrow[\text{galactokinase}]{\text{inducable}}$ Gal-1-PO4 + ADP

E_4

$E_4 \sim Gal$
|
T $\sim$ UDP

Gal-1-PO4 $\xrightarrow{E_X}$

Gal-1-PO4 + UDP-Gl $\underset{E_3}{\overset{\text{inducable}}{\rightleftharpoons}}$ UDP-Gal + Gl-1-PO4

$E_3 \sim Gal$
|
T $\sim$ UDP

UDP-Gal

Gal + ATP $\longrightarrow$

5

Then ~~consistent and~~ compute it
in anamnestic response on assumption
until r_0 is $1/2$ day post injection.

When bug grows on Gal and no glucose present
UDP-GAL low and E_3 as well as E_4 must be induced
When grown on glucose UDP Gal high (b) (because of
only escape cell wall) and E_3 & E_4 synthesis repressed

4

β-Galactoside Story

$$Gal + ATP \xrightarrow{\text{gal-kinase}} Gal-1-PO_4 + ADP$$

E_3

$$\boxed{\begin{array}{c} E_3 \sim gal \\ I \quad \quad \Big| rep \\ T \sim UDP \end{array}}$$

$$Gal-1-PO_4 \xrightarrow{E_6} cell\ wall$$

$$Gal-1-PO_4 + UDP-Gl \xrightleftharpoons[E_3]{\text{uridyl transferase}} UDP-GAL + Gl-1-PO_4$$

$$UPDGAL \xrightarrow[E_5]{\text{β-galactosidase}} UPD-GAL^*$$

$$\boxed{\begin{array}{c} E_5 \sim gal^* \\ I \quad \quad \Big| rep \\ T \sim UDP \end{array}}$$

$$\boxed{\begin{array}{c} E_3 \sim GAL \\ T \quad \quad \Big| rep \\ \sim UDP \end{array}}$$

$$UDP\_GAL \xrightleftharpoons[E_2 \ (slow\ diss \leftarrow)]{\text{epimerase}} UDP\_Gl$$

$$UDP-Gl + PP \xrightleftharpoons[E_1]{\text{pyrophosphatase}} UTP + gl-1-PO_4$$

$\boxed{E_5}$ $\Big\}$

$$\begin{array}{cc} E - TMb & E^+ \ of\ Gal \\ I & \\ T & T^* \ of\ UDP \end{array}$$

UDP Gal-r

Harriman with Spiegelman - Meyoth

β-galactosidase

10

... may play an important

... role in regulating

the ~~true~~ level of enzymes ~~which~~

catalyse the production of

which lie in the pathway

of the synthesis of ~~possible~~ ... such

such as for instance amino

acids and that ~~perhaps each~~ intermediate

metabolites introduce the enzyme

that catalyses its ~~next biochemical the next step~~

~~step along the setting~~ ...

... of the biochemical pathway ...

... the terminal

thus one may have thought

that in the synthesis of ~~arginine~~

~~along the pathway~~ leading from within

... ornithine

+ acetylornithine E_3 ornithine E_2

ornithine E_1 arginine

acetylornithine induces E_3 the

formation of E_3, ornithine induces

the formation of E_2 etc in the case of E_3

& Vogel has shown that this

is not ... the case, but rather

... that Arginine represses the

formation of E_3, and he has stressed

~~Chemical Analog theory of enzyme~~
~~repressor mutations~~ ~~suppressor~~ induction
~~Suppressor mutations~~ ~~and~~ adaptation
~~Metabolites and adaptive enzymes~~
~~Enzyme Adaptation.~~

Adaptive (in modern microbiology)
are ~~certain~~ (enzymes) ~~which are~~
normally present in small amounts
in growing bacteria but which may
~~but are~~ ~~or are which~~ appear
greatly increase in amount of the
bacteria grow in the presence of
certain ~~bacterial~~ chemical
~~analogs~~ compounds which the bacteria
can metabolize ~~or their chemical~~
~~analogues~~ are called inducible enzymes
~~and the same~~ chemical compounds which
make their ~~by~~ increased synthesis are
called inducers. ~~It was generally~~
~~believed~~ Recently largely due to the
work of J. Monod and ~~his labor~~ at his
(others)
laboratory much attention was pressed
on the phenomenon of enz. ind
in general and the enzyme β-galacto-
sidase (which splits lactose) in parti-
cular ~~Up until~~ ~~repeated~~ ~~it has been~~ recently
~~has been surmised~~ it has been generally
surmised
It was generally believed that
the phenomenon of enzyme

personally
~~to be approached~~

It find it indeed difficult to think
of a mechanism by which any
of the ~~alleged~~ known "inducers" could
through interacting with a template
induce that template to an earlier
production of the precise
enzyme it is making; therefore
I set out to see if Mass's
postulate can be substantiated
in force of the apparently over-ruled
ming evidence to the contrary

The present paper is a summary
of my conclusions. — In order
to justify the concepts with which
I must operate) I must mention
and to arrive at a
correct formulation of the problem
I must start out with some of
the conclusion reached and the assumptions which
to reach them enabled us
1.) ~~There are~~ as ~~stated~~ ~~as~~ ~~and as~~
~~indeed~~ favorable factors —
~~intermediates or interm~~ as well
as certain intermediate
~~metabolites~~ can repress the formation
of enzymes as illustrated in the

[~~and accordingly organize~~ matters ~~then~~ be repressed]

that there is no indication that

~~an~~ inducers operate in this system

Werner Maas ~~and H above and~~
~~priority demonstra~~ ~~an extension~~ and fried
in ~~investigations~~ ~~at the appearance~~
for the ~~investigated~~ ~~The dependency the enzyme~~
for the ~~same~~ of the ~~level of the~~ ~~amount of E~~ present
in the ~~growing~~ bacteria ~~as a~~ in the
bacteria on the presence ~~H~~ or absence
of organine, mutation and mutation
and could find no indication for

any "induction". ~~By induction~~

~~we mean little~~

Werner Maas reviewing ~~little~~ ~~other~~ several other
~~second~~ similar biochemical
pathways and finding ~~size~~ a similar
situation finally ~~postulated~~ in every case
~~postulated~~ reached the
~~conclusion~~ that inducers ~~in~~ the

sense of compounds which enhance

the synthetic rate of an enzyme

by interacting ~~at~~ with the Template

which synthesizes ~~could~~ the enzyme]

~~may~~ not exist.* ~~If true~~ ~~by this discovered~~
~~and~~ If this be true ~~we~~ must find
some explanation for the apparent
existence of inducer, ~~and find this~~

* ~~But even~~ And even

Berkeley Nov 27/57

while the suppressor is attached
no enzyme is made. [crossed out: In general as such supp is more]
Generally such suppressions - for we assume -
must is always, specific and
It occurs only if the compound
complexed to R (since the universal
carrier as one of several possible
universal carriers) has specific
affinity

Having thus defined what a
suppressor [crossed out: is] compound is we may
now [crossed out] define what an real-
[crossed out] as opposed to the rather loosely
defined term of inducer [crossed out: for every]
which is used whenever a substrate
evokes the appearance of an enzyme
[crossed out: As an exexample for such an
inducer which] We shall speak
of a real inducer only if we
deal with a compound that inhibits
an enzyme that produces a suppressor
because it is a chemical analog of the mistake
[crossed out: As example for an inducer of that enzyme inhibit]
In general the real inducer [crossed out]
is therefore a chemical analog

case of arginine the repressor
need however not be the growth
fact, arginine itself, but more likely
it is a deduction produced by an
enzyme (ER) which need not in the biochemical which is
pathway leading to the end product in our example as the end
product. This enzyme couples a ~arginine)
non specific radical R to arginine
and the repressor is R-arginine
In the case of kynurenic acid the
arginine the suppressor may
well be the Adenyl kynurenic acid.
Suppression of the an enzyme of the
biosyn that lies in the bio-
synthetic pathway of any
occurs according to this view
by the reversible union of
the enzyme template complex with
the super suppressor which may by
be syn inactivated by the specific

enzyme template complex $\{ E \sim AA \atop + \sim R \}$ suppressor and gives the
when E ~ AA
while)

the R-template complex is free +~Adenyl
the template makes enzyme at the
full maximal rate.
super

But when the cell grows on glucose the
UDP Gal becomes high may reach
a high level for it only function
too is probably to supply galactose
for the synthesis of cell wall. —
Ytherefor thus galactose only one
might say if one wishes that
it "induces" the formation of E₃
is not a ~real~ inducer in the term of
our definition. In
a more consistent terminology one
would say that all carbon sources
except galactose are inhibitors
of enzyme E₃ . A real inducer of E₃ would
by an inhibitor of E₂ and there β-gal
So from our point of view
it is useful to divide the real inducer

a suppressor into two classes

1.) inducers which
are chemical analogues
of a substrate of an enzyme that
makes a suppressor compound and
therefor inhibitors of that
enzyme

(which are known
which induce galactose
an enzyme which to induce the large β-gal

of the substrate of the
enzyme that produces the suppressor.
An example for an inducer, i,
that is not a real inducer γ, is
galactose & the metabolic pathway
in this case was described by K and
is as follows:

E_3 and E_4 are "inducible" enzymes
in the sense that when the cell
growth on galactose they are
present at a high level while
when the cell grows on glucose
there level is very low. We assume
that UDPGAL is the suppressor for
E_3 as indicated by the symbol

$$E_3 \sim Gal$$
$$\dashv -UDP$$

When the cell grows on Galactose
UDPGAL need not accumulate to a high level
for the end product UTP and since
Gl-1-P04 are needed as an energy source and as
a carbon source by the growing mechanism.

suppressor

An example for this ~~first~~ class ~~is probably~~ ~~one of the best~~ ~~studied system~~ — the inducers.

~~E₂ When grown on~~ the bacteria are ~~unable to~~ metabolize or ~~break~~

~~the induce~~ β-galactosidase can be induced by β galactose ~~a number~~

of β-galactosides ~~which presumably inhibit E₂~~

(β Jacque Monod and his co-workers have shown that ~~the inducers interact~~

and his co-workers

~~In this class fall~~ ~~presumably the~~ numerous — the

numerous ~~as found by J. Monod~~ β-galactosides ~~interact~~ induce

β-galactosidase. They presumably

~~induce~~ ~~inhibit~~ E₂ ~~attack~~ and thereby lower the ~~above~~ conc. of ~~Gal~~ UDPGal

which ~~is the key~~ may be the repressor

of the repressor

(or a precursor ~~element~~) of β-galactosidase.

~~Were in this paper man~~

FMG ~~if~~ ~~of~~ ~~indeed inhibits~~ E, as I suspect it does — would fall into this class

1.) Those which inhibit an enzyme (because
that lies ~~on the~~ always one ~~along the pathway~~ these are
of the normal metabolic pathways ~~of~~ ┐ chem
the ~~bacterial~~ cell or produces a suppressor ⎿ analog
(or a mirror of a suppressor) ∫ of

2.) Those which ~~inhibit~~ an enzyme
that produces a suppressor, ~~for another~~
~~enzyme~~ the suppressor might ~~say~~
~~suppress the formation of an enzyme~~
~~that lies in along one of the metabolic~~
~~pathways but in general~~
~~in general that suppresses in general~~

~~not~~ an enzyme that acts on an ~~endo-~~
~~enzyme~~ ~~but in~~ but ~~along~~ does not lie
along the metabolic pathway. ~~In any~~
~~or may be any~~ normal The general ~~case is illustrate~~
~~in table I~~ ~~the reactions~~ For the
sake of argument ~~one~~ ~~way~~ we may
~~assume~~ that S_0 denote an ~~amino~~
acid but it can't ~~not~~ feed ~~denote~~
any growth factor that the ~~bact~~
can make use of. S_1, S_2, S_3 etc
~~are successive intermediates~~
S_1 is one ~~metabolic~~ step removed from
S_0. Similarly S_2 is one ~~metabolic~~
step ~~removed~~ from S_1 etc

~~Enter this paragraphs~~ ~~ends~~ ~~with~~ S_n, T_0,
~~with this second class of~~
~~indices.~~
                              proper

is compiled for R. Only then can
we understand why there are so many dullness
y which can find an I_n
for which they resemble
sufficiently to be able
to inhibit the enzyme E_n

$$(R - \mathcal{I}_n) \xrightarrow{E_n} (\mathcal{I}_{n+1} - R)$$

$\mathcal{I}$

What can be the evolutionary origin of such
a large variety of enzymes and the
repressor R which

It is our contention This paper

In the following I shall
describe one way in which
they may arise

let us focus our attention
on one template which forms
an enzyme that couples a
new factor $\frac{y_0}{o}$ for instance an amino
acid such as tryptophane to a
carrier R to yield the repressor
$(\mathcal{I}_0 - R)$

paper

$$\text{precursor of } Y \quad \overset{\large 28}{\underset{\equiv}{}} \quad Y \qquad Y+R \quad \overset{\equiv}{} \quad (Y-R)$$

We propose to deal in this paper
mainly with this second class of real
inducers. there is a ~~great number~~ ~~innumerable~~
compound which bacteria can
~~oxidise~~ and metabolise. ~~the~~
~~some~~ these compounds do not resemble
closely any of the ~~normal~~ ~~usual~~
~~known metabolites~~ ~~products of~~
no intermediate which lie on
the direct path of breaking down.
Yet ~~the bacteria exposed to this~~ these
compounds ~~by least~~ make induce
enzymes which can carry them through
one or more biochemical steps.
If they are are real inducers in the
~~by self~~ ~~there that the metabolic~~ sense
of our definition ~~and ~~have~~ why~~
~~are these~~ ~~does the bacterium~~
the bacterium must contain a
~~long must be~~ a very great variety
of enzyme produce ~~repr~~ suppressors
~~all turning the~~ ~~so~~ many of them
having the same carrier R but
differing in the compound Y on which

proper

~~We will designate this by the symbols~~

~~E₀*~~
~~T*~~

We ~~will~~ designate the altered template and enzyme with E_0^* and T^* and express ~~the fact of~~ the weakening of the bond by writing:

$$E_0^* \not\propto T_0$$
$$T^* \not\propto R$$

~~By~~ Because of the weakened bond the enzyme-template complex will ~~now~~ be free a greater proportion of the time and will the produce more E_0^* ~~and~~ The level of $(E_0 - R)$ will then be raised less ~~so~~ ~~enzyme~~ E will be produced and less T_0. If the change is large enough the weakening will be ~~catastrophic for T_0 and will~~ unable to grow unless T_0 is ~~supply~~ supplied.

This can be remedied by a repressor mutation occurring in some other gene which provides a template that ~~synthesizes~~ forms an enzyme E for which we have ~~super~~

30

The growth factor Y is ~~produced~~ reduces produced from a precursor by an enzyme. And the ~~suppressor~~ permabrings the ~~required~~ ~~level~~ to the proper level. We may write:

$$\text{Precursor of } Y_0 \xrightarrow{\ E\ } Y_0 \qquad R + Y_0 \xrightarrow[\substack{E \text{ or } Y_0 \\ + \sim R}]{\ E_0\ } (Y_0 - R)$$

$$E = Y_0$$
$$\overline{T} = -R$$

Y also ~~keeps~~ holds ~~down to the proper~~ down the rate of its own synthesis to the proper rate

We shall soon consider what happens ~~if~~ ~~a mutation~~ a mutation of a certain type occurs in the template that makes the ~~suppressor~~ Enzyme E_0 that produces the ~~suppressor~~ ~~if~~ and if this mutation later on reverts ~~and subsequently~~ and then keeps on occurring and reverting.

~~With~~ We have in mind a mutation in the template (that may or may not ~~reflect~~ affect the enzyme itself that makes the ~~band~~ between the enzyme template complex and the ~~suppressor~~ compound

proper

Later mean another mutation from
T to T* occurs a further suppressor
mutation of the same general
types may correct it so that
we obtain a series of enzymes
where

$$(R - y_{n-1}) \xrightarrow{E_n} (y_n - R) \quad (R - \tfrac{y}{n}) \quad \text{and } t_n$$

y_n is n steps between steps
away from the original amino
acid (or other essential
metabolite) and may bear
little chemical resemblance
to it.

But in this general

$$
\begin{array}{ll}
E_n \sim y_n \\
\dagger \quad | \\
\quad \sim R \\
E_n \longrightarrow y_{n+1} \\
\dagger \quad \sim R
\end{array}
$$

$$ y \to E_n \to y_{\text{mod}} $$

This may we may understand

Because of the abundance
of

proper

$$y + R_0 \xrightarrow{E_0^*} (y_0 - R) \qquad (R - y_0) \xrightarrow{E_1} (y_1 - R) \quad H$$

$$
\begin{array}{c}
E_0^* \sim y_0 \\
T^* \sim R
\end{array}
\qquad
\begin{array}{c}
E_1 \sim y_1 \\
| \qquad | \\
T \sim R
\end{array}
$$

and
$$
\begin{array}{c}
E_0^* \sim y_1 \\
| \qquad | \\
T \sim R
\end{array}
$$

This suppressor mutation now by producing the inhibitor $y_1 - R$ that is sufficiently strongly bound to the enzyme template complex $E_0^* - T$ corrects the defect of T^*

If there there is now a back mutation from T^* to T, then there is again an imbalance which needs correction and there may occur a second suppressor mutation producing an enzyme E_2 so that we may have

$$(R - y_0) \xrightarrow{E_1} (y_1 - R) \qquad (R - y_1) \xrightarrow{E_2} (y_2 - R)$$

$$
\begin{array}{c}
E_1 \sim y_1 \\
| \qquad | \\
T \sim R
\end{array}
\qquad\qquad
\begin{array}{c}
E_2 \sim y_2 \\
| \qquad | \\
T \sim R
\end{array}
$$

$$
\begin{array}{c}
E_1 \sim y_2 \\
| \qquad | \\
T \sim R
\end{array}
$$

Proposed

35

for proper,
Plastic & volume forces
authority

or that the template is free than the repressor is dissociated off than
the template makes enzyme at the
full rate.

$$(R - y_{m-1}) \frac{E_m}{\quad} \quad (y_m - R) \quad (R \cdot y_m \xrightarrow{E_{m+1}} y_{m+1} - R)$$

$$\begin{array}{cc} E_m & y_m \\ | & | \\ T \sim R \end{array}$$

$$\begin{array}{cc} E_{n+1} & y_{n+1} \\ | & | \\ T \sim R \end{array}$$

When a compound y is presented to
the bacterium which is a chemical
analog of y_n the enzyme E_n may catalyse
a similar biochemical step.

$$y \xrightarrow{\quad E_n \quad} y \text{ unadsorbed} \qquad E_{n+1}$$

Such a compound will act as an inducer
by competing for the enzyme with the
repressor $(R - y_{m})$ and also by inhibiting the
enzyme E_{n+1} which also produces the
repressor and thus decreasing the output
of the repressor $y_n - R$. thus there will be
an increase in the synthetic rate of E_n.
There may be many series of perhaps
10 to 100 members each, and each
series may originate with a repressor
composed of an essential metabolite coupled
to a carrier. In the case of RH for
instance the first member could
be an Adenylanurno acid in each series.
As we proceed to a higher number of
an along one series in each biochemical
individual step (which leave the carrier R

In a proper dealing with the
the formation of ~~inducible~~ in bacteria
~~enzymes~~ and other induced
I have endeavoured to show that
it is possible to understand ~~the~~
~~response of~~ why bacteria respond
to a great variety of inducers. ~~This~~
My explanation ~~was~~ based on
the presence in bacteria of a great
variety of enzymes E_n each of
~~inducible~~ ~~act~~ ~~can repressors~~ ~~from~~ that
~~can reversibly combine the enzyme~~
~~template complex that synthesises the~~
~~enzyme. these repressors are small~~
~~are compounds of small molecular~~
~~weight~~
that it transforms. The repressors
are compounds of small molecular
weight of the form $(I_n - R)$ where
R is a universal carriers ~~probably~~ [In the
one
of several
case of A.A the carrier is presumably
an adenyl amino acid] ~~the~~ the repressor I_n
can reversibly combine with the template
— Enzyme complex ~~of~~ which produces the
~~template~~ enzyme E_n and probably also
with the enzyme E_{n+1}. ~~As long as~~ the repressor While
is attached to the
enzyme template ~~complex~~ complex
the template does not make enzyme

T Proper

enhances the rate of formation
of the antibody. There remains
not provided [crossed out] the theory does [crossed out] that give a model
of the how this enhancement is accom-
plished by the antigen. Nor are the
basic phenomena — such as the greatly
shortened latent period — [crossed out]
the second injection of an antigen
[crossed out] the and the greatly [crossed out] rate of
antibody production [crossed out]
[crossed out] that follow the second
injection if (provided [crossed out]
[crossed out] there is an inter-
val of several week between
the two injections) [the phenomena of
the toxic phenomena]
[anamnestic]
response and tolerance [crossed out]
[crossed out] explainable by this
satisfactorily explainable on the
basis of the theory. In the anamnestic
response [one] upon the second inject
[the response]
the anamnestic response is obtained
If [one allows] subsequent to a first inj. at an antigen
one allows several weeks to
pass and then injects the same
antigen a second time. antibody
then appears after a latent period
which is ⅓ to ⅒ of the latent
period of [³/₄] the first injection
and the rate of antibody prod
may [be] more than 10 times greater
than after the first inj.
[crossed out]

~~unchanged~~ ~~carries the repressor~~
~~further and~~ further away from
the original changes the chemical
structure more and more until
there is no ~~remembrance~~ left to
the original amino acid moiety.
~~In the~~ I have ~~failed to make it~~ ~~plausible~~
~~in what manner~~ this great ~~variety~~
~~of repressors~~ this variety of repressors
arose — according to my theory — through
repressor mistakes which compensate
for mistakes ~~of the other of~~ which
affect the amount of the amino acid
or the corresponding adenyl amino acid.
~~My things above~~ ~~Amino~~ / ~~I was speaking here~~
~~of amino ac~~ I have singled out
amino-acids over other essential
metabolic ~~only~~ here only for the sake
of convenience of argument.

┤ ~~Introduction~~ 1/11

A Chemical analogy theory of
antibody formation —

~~The~~ ~~It is~~ apparent that the
Ehrlich-Jerne theory of antibody
formation assumes that all antibodies
that say a Rabbit can form in
response to the injection of an antigen
are formed in ~~for example~~ ~~small~~
about amounts ~~by the rabbit~~
of ~~ever prior to~~ ~~being repressed~~ first
injection of the antigen and that
~~somehow the antigen~~

(4) and that when the antigen reaches the antibody template complex it combines irreversibly with it and thus causes tolerance.

The theory here presented is based on the assumption that some cells of the lymphatic system produce a great variety of enzymes E_n which produce such a specific compound of small molecular weight. Many of these compounds have one part in common the carrier R — whole by ~~differ but the~~ father a part designated by H_n is. the compound which may thus be described by the symbol (H_n - R). ~~is a repressor~~ ~~fourth four~~ the enzyme E_n catalyses ~~as a repressor. It can reversibly combine first of all with the enzyme template complex and the combine~~ the reaction.

$$(R - H_{n-1}) \xrightarrow{E_n} (H-R)_n$$

(H_n - R) is a repressor of the ~~rate~~ production of the enzyme E_n. ~~The~~ It acts as a repressor by reversibly combining with the template that synthesizes the enzyme E_n or to be more precise

The Ehrlich ferme theory seems
to have some elements of truth in
it but it contains only half of
the truth; This paper ~~proposes~~ ³ᵈ ~~belief~~ —
to supply the missing half of
the theory. It should therefore
~~and~~ enables us to
understand the basic phenomena
of antibody formation and
~~makes~~ predict the outcome of
certain hitherto not performed
experiments in principle

It assumes like the Ehrlich ferme
theory ~~that the all antibodies that~~
~~flow~~ ~~the too begin~~ the antibody
~~forming~~ cells of the lymphatic tissues
of the rabbit ~~its~~ capable of making
an antibody, have a template which
is specific for that antibody.
and that prior to the injection
of the antigen ~~in the~~ the formation
of the specific protein is repressed
and that upon of injection of an
antigen that ~~contains~~ has a hapten
that can specifically combine
with the ~~antigen~~ antibody this repression
is temporarily lifted, but here
the similarity ends. I do not assume
that the antigen reaches the ~~template~~
specific antibody — template complex
in the adult Rabbit which can
form antibody. On the contrary

I assume that it can reach it
only in the ~~jaw of~~ newborn Rabbit
which can not form antibody

43   While the antibody template
complex is thus covered by
a repressor no antibody is formed
by it. When the repressor
is dissociated off antibody
is formed at the full rate ~~as~~ ~~stated~~ ~~before~~ ~~while~~ ~~without~~ ~~the~~ ~~duty~~ ~~of~~ ~~any~~ ~~repressor~~ ~~can~~ ~~the~~ ~~antibody~~ ~~injected~~

In contradistinction to the
~~an~~ antibody-template complex
~~which~~ ~~is~~ ~~the~~ ~~antigen~~ ~~or~~ which
~~neither~~ ~~adult~~ ~~rabbit~~ the antigen
~~can~~ ~~not~~ ~~reach~~ the Enzyme-template
complex can be neutralized gradually the antibody
~~but~~ ~~it~~ ~~can~~ ~~be~~ ~~reached~~ ~~only~~ ~~but~~ ~~per~~
~~slowly~~ ~~and~~ ~~with~~ ~~difficulty~~
When an antigen ~~coming~~ ~~in~~ happens to
reach the ~~anti~~ enzyme template
complex ~~with~~ which is specific for it
since it is a chemical analog
of the there is an immediate combi-
nation formed

$$
\begin{array}{ccc}
E_n & - & H \\
| & & | \\
T & - & Prot
\end{array}
$$

Further upon dissolution of the antigen
all the enzyme E_n which is present
in the cytoplasm of the cell that makes
this enzyme is reached by the antigen
$Prot-H$ and the irreversible complex
$$E_n-(H-Prot)$$
is formed.

with the enzyme template [42]
complex as is described below

$$\text{enzyme template} \begin{Bmatrix} \frac{E}{\text{complex}} \sim H \\ \frac{1}{T} \sim R \end{Bmatrix} \text{repressor}$$

While the enzyme template complex
is thus covered the template
can not form enzyme. When the
repressor is dissociated off the
template makes ~~way~~ ~~together~~
~~enzyme~~ enzyme at the full rate.
~~The enzyme~~ ~~at~~ For the sake of
argument we shall regard
the antigen as composed of
a nonantigenic Protein and a hapt
ten H. Thus antigen = (Prot - H)
If the hapten H is a chemical
analog ~~to~~ of some group H_n of
a repressor ~~contained in the~~
represented in the rabbit
then that repressor will also
reversibly combine with
with that ~~antibody~~ antibody
~~template complex that~~
forms an antibody that is
specific for hapten H

$\overline{H}$ repressor

45

If We ~~have~~ assume that
the repressor (Hn-R) are ~~inhalatoxes~~
in the rabbit with a half life of
the order of a day and that the enzymes
E ~~n are~~ have a half lives of a ~~wee~~ few
weeks or months. ~~we can now~~
~~predict the sequence of events in~~

~~X·X·X~~ Fig illustrates !

We shall now ~~describe~~ ~~deserve what~~ the
~~repeatedly~~ response of the rabbit ~~the~~
to the injection of an antigen on the
basis of some plausible assumption
of the quantities involved

We shall assume that in a non pre-immunised
animal the titer of the repressor has to
drop 100 fold before there is enough
antibody-template complex freed from
repressor to give an appreciable
antibody production. ~~the one~~
~~of the~~ neglecting for the moment any
the fact that ~~ten~~ ~~ratio~~ ~~of the enzyme template coupled is~~
~~it ~~ ~~absolutely irreversibly~~
~~only ~~ ~~will be irreversibly tied up~~
~~by the antigen~~ gradually + irreversibly
~~tied up~~ of the antigen. We can then
immunoite

47

Gross repressors - differentiation
and the process of aging in mammals

III Paper

49

small number of
Templates for t₀
Same cells will produce much
antibody

antibody tumes

cells divide and templates
are produced again. —

If enzyme lives longer in one line long
most one dies into body prod continue
in ? serum Rabbit

Secondary response
differs from primary
because T-E does not fall
much !

Can you protect with trypsin
so that first day of antigen
does not cause anamnestic
response
Can you cause tolerance
by injecting trypsin with small
protein ?

Antibody

Explain whole
minimum local curve
exists.

If A-E is destroyed in first injection
but little destruction in second observed
shape may be explained

Write rate of antibody per template

$$N(m) \; a_m \; \frac{1}{1 + \frac{x_m}{k_m}} \quad \text{is rate of Antibody prod}$$

x_m conc. of inhibitor ; k equilibrium
constant

assuming $t=0$, $x_m^{(0)} = 10000 \; k_m$ Kinends

$\downarrow x_m^{(1)} = 100 \; k_m$

$x_m^{(2)} = 50 \; k_m$

Titer spectrum
$a(m) N(m)$

Enzyme asymptotic over :

$\frac{E}{I}$	$\frac{E\sim gal}{I}$	$\begin{array}{c}E\sim gal\\ \mid\\ I\sim UDP\end{array}$

What is (true) inducers and pseudo inducers in enzyme induction ?

$$x \cdot e^* = K_1 [xe]$$
$$y \, e^* = K_2 [ye] \qquad 1 = e^* + K_1 [xe] + K_2 [ye]$$

$$e^* = \frac{1}{1 + \frac{x}{K_1} + \frac{y}{K_2}}$$

$$e^* + [xe] = \frac{1 + x/K_1}{\left(1 + \frac{x}{K_1} + \frac{y}{K_2}\right)} \qquad [ye] = \frac{y/K_2}{1 + \frac{x}{K_1} + \frac{y}{K_2}}$$

rate of eng proad proportional

$$\frac{\dfrac{1}{K_2}[UDPGAL]}{1 + GAL}$$

$$\frac{1 + [GAL]/K_1}{1 + \dfrac{[GAL]}{K_1} + \dfrac{[UDPGAL]}{K_2}}$$

rate of break up

$E \sim$

better for galactose

assume precursor x, amplitude y,
what synthesis of E

$$\int \frac{d}{dt} c = \boxed{\frac{1}{1 + \frac{y}{k_R}}} - \frac{a}{c} \qquad \sigma = 1$$

Amp.

$$\frac{dy}{dt} = x e - b y - y$$

$$b + y = x e$$

$$y + by = x \frac{1}{1 + \frac{y}{k_R}} - b$$

precursor

$$y + \frac{y^2}{k_R} + b \left(1 + \frac{y}{k_R} \right) = k_R x e$$

do this with amplit & interhibition
at template level! Amplrine complex
with $(AA-R)$

template complex
$$AA + R \xrightarrow{\;\;y\;\;} (AA-R)$$

To Molt: What about proving cells are protective
53 and study THG and all other influences?
May 4/57

Jeans: repressor ~ repressor
antagonist
His kind of adaptation always less
sharp than tumor. —

K_r for UDP GAL   might be

assuming 1 mµµ/cc (10^{-5} molar)

$\Delta M = RT \log K$ and $\dfrac{UDPGal}{K_r} = 1000$

$K = 10^{-8}$ mol

$8.2.3 = 16 \dfrac{24}{18.0}$ $\boxed{\text{col 1000}}$

~~~~~~~~~~~~~~~~~~~~~

Kalckar proposes   Kiyoshi Kurahashi
Science Vol 125   pp 114 & 116 1957
Arthur Landskron   (the first two enzymes
                          E4 and E3 are not
                          not present when
                          cell (col) is grown
                          on glucose in absence
                          of Gal but present
                          when grown in presence
                          of Gal

Loous: J.E. Med. 104 (1956)

~~~~~~~~~~~~~~~~~~~~~

Stanier R.Y., Simultaneous adapt.
      J. Bact 54 339-348 /47

<u>Antibodies:</u>

Caries. — cells in here / all samples taken to
added milk have much antibody

Orham Frank J. et al Maurer, Paul H. Wolfe
and Berkmuller Maria P.   p425 Vol 103 1956

Rabbit + brain / flab

                       myg persistent / antibody

Evans B. exp Med 102   p49. 1955 brd
Evans Albert H, Hearn, Elisabeth   61 { rabbit Tn
                                        73 { tetanic respon
    rabbit ?   83.   -

$$\frac{dr}{dt} = P E_0 - \frac{r}{\tau_r}$$

$$\boxed{\tau_r = 1 \text{ day}}$$

$$r_{t=0} = P E_0 \tau_r \quad \text{for } t=0$$

$$E_{\prime\prime} = \frac{E_0}{f_1}\left(1 - e^{-\frac{t}{\tau_e}}\right)$$

$$\boxed{\begin{array}{l} \tau_e = 10 \text{ days} \\ f_1 = 100 \end{array}}$$

1.) $$\frac{dr}{dt} = \frac{P E_0}{f_1}\left(1 - e^{-\frac{t}{\tau_e}}\right) - \frac{r}{\tau_r}$$
$$\boxed{r = P E_0 \tau_r \quad \text{for } t=0}$$

2.) anamnestic response no change size

$$\frac{dr}{dt} = \frac{P E_0}{f_2}\left(1 - e^{-\frac{t}{\tau_e}}\right) - \frac{r}{\tau_r}$$

$$r = \frac{P E_0 \tau_r}{f_1 \tau_r} \quad \text{for } t=0$$

$$\boxed{\begin{array}{l} f_2 = 100 \\ f_1 = 100 \end{array}}$$

write P for A above!

$$\boxed{\text{write } X_m \text{ for } r}$$

$$\frac{dA}{dt} = N(m)\, a(m) \frac{1}{1 + \frac{X_m}{k_m}} - \frac{A}{\tau_A}$$

$$\boxed{A = 0 \quad \text{for } t=0}$$

$$P E_0 \tau_r = 10^4 k_m$$

$$\boxed{\tau_A = 10 \text{ days}}$$

put $P E_0 \tau_r = 1$
$$N(m)\, a(m) = 1$$

$$\Delta F = -RT \ln K$$ $$\Delta H = -RT \ln K$$

cellubioronic acid as hapten
makes antibody reacting with type III and VIII
III and VIII contain this hapten   polysaccharides
cellubioronic acid induces (some have)
an enzyme which depolimerizes III and VIII

→ assume molar conc. of antibody

$$\frac{1 \; \mu \; gm / liter}{10^{-8} \; mol / liter}$$

K   • Enz

∞ N⁻   10⁻⁶

$10^{-16} \quad 6 \times 10^{12}$ molecules 10^{5} cm

58 phenomena ~~on the shores~~ on the
basis of a ~~common produced pts~~
the ~~control~~ mechanisms ~~~~
that ~~appear~~ to control the level
of enzymes in cells.

These ~~are~~ seemingly unrelated
phenomena are as follows

b) ~~adaptation to galactose~~ of ~~Backwards~~
~~when~~ adapt to galactose
~~grown~~ in minimal medium
~~(glycerol)~~ ~~a~~ in the presence of galactose

a.) ~~the fact that~~ ~~the fact that~~
~~there are a number of suppressor~~
~~mutations in neurospora which~~
~~restore~~

There are a number of different
mutations in neurospora all lying
within the same functional gene
which lead to ~~less (~~ or the ~~reduction~~
absence [~~or~~ a greatly reduced level] of the
enzyme ~~tryptophane synthetase~~
[~~the~~ an enzyme plant condenses indol
undersin ~~and~~ to give tryptophane.

[wrote ~~Jadawfly~~] ~~The ability to ~~provide too~~~~
~~occurring in various different time~~ (suppressors) can
2.) The fact that ~~these~~ there is
a great variety of compounds of
small molecular weight which
do not appear to be normal
metabolites at ~~bacteria~~ . Yet
~~bacteria~~ can metabolize them
~~and~~ ~~respond~~ to them in many case

enzyme Detour here

~~attempts in the name~~ and I ~~at once~~ made ~~a~~
~~an~~ perfunctory examination
~~and~~ of the known poets of the
in Duchovne of the enzyme to
On Duchovne dose ~~made~~ ~~the~~ which me
at once me which me ~~jointly~~
~~undisturbed~~ which from this point
of view ~~formulated by terms~~ which
~~another theirer took~~ they ~~must~~ ~~be~~
~~at once~~ looked on the base of it
encouraging.

~~In the following I shall attempt~~
~~to interpret a number of apparently~~
~~puzzling phenomena~~ which are
~~not obviously~~
~~In the following I shall attempt~~
~~to~~

(we have)

— Maas ~~XXXX~~ expressed the hope that
whenever ~~there is~~ a case of apparent
enhancement of the production of an
enzyme ~~a me it might turn out~~
~~be~~ in the end the interpretable as the
~~an inability to an inducer than of the~~
~~production~~ ~~in inhibition~~ ~~ession~~ of the production
~~of an inhibition~~.
                                          set forth
I shall attempt to ~~develop~~
in the following a set of assumptions
which ~~of interlinked~~ may
permit us ~~to interpret~~ to
interpret a number of important
— and seemingly unrelated ~~paper~~
   I paper detour   paper; detour

an adenyl amino acid.

(with E_2 and free 2 AA → also free 3 AA)

$$\downarrow \text{free } AA \xrightarrow[E_1]{} AA \xrightarrow{E_a} \text{adenyl-}AA +$$

$$\boxed{\begin{array}{c} E_1 \sim AA \\ | \\ T \sim R \end{array}} \qquad \left| \begin{array}{c} E_0 \sim AA \\ | \\ T \sim R \end{array} \right|$$

_____

Need mantle an inducer operate ?
No point of leaving an inducer
Galactose story   Principle

H. J. Vogel 140 (23) 145 Proc. Nat. Ac. Sc. US 39 $\frac{570}{53}$
    Charles Yanofsky p. 147 & Enzymes
units of Biological   structure and
function.   the formation of the amino acid

the enzyme E_1 that catalyzes the formation are synthesized along-
side their   specific template T we may
We assume that   and assume that
the enzyme   there will be an acquilibrium
or at a stationary state—   reaction—according
in which the enzyme template complex

T–E is reversibly combined with
the AA, a precursor of AA and or adenyl AA
we may   to form

$\begin{array}{c} E_1 \\ | \text{ or} \\ T \end{array} \quad \begin{array}{c} E_1 - \text{Prec } 3AA \\ | \\ T \end{array} K_{prec AA} \text{ or } \begin{array}{c} E_1 - AA \\ | \\ T \end{array} K_{AA} \text{ or } \begin{array}{c} E_1 - AA \\ | \\ T - \text{Adenyl} \end{array} K_r$

We shall now further assume that
           the template are not form
the enzyme while "coupled" with
adenyl AA but can

3 the fact that ~~was~~ establish —
and ~~many~~ many other androids of
course can form ~~only~~ ...

5) ~~...~~ certain
a ~~...~~ phenomena ~~...~~ your
cooperation

~~For the art~~ In order to make
~~communication easier~~ I shall
~~make~~ For the sake of ~~the order~~
~~clarity of my~~ making connections ~~easier~~
I ~~shall proceed~~ make my assumptions
more ~~...~~ detailed and ~~...~~ (and therefore
less likely to be true) ~~in all of it~~
then ~~conceptual~~ than is demanded
by the ~~real answer~~ thoughts that lies
behind the argument. I believe however
that the mistakes I may thereby
introduce can be easily corrected
later on —

~~Thus for instance~~ First of all I
shall make a ~~specific~~ ~~assumption~~
~~two~~ concerning the way an amoeba
~~to~~ avoid such as amoeba represses
the formation of ~~one~~ an enzyme
that lying along the ~~...~~
pathway leading from some precursor
~~to the~~ through several ~~...~~ steps
to the AA itself.
I assume that the AA itself is
not the repressor of ~~the~~ any such
enzyme and that there is beyond
the AA itself another step that yields

It is not clear why the spermatozoa (or
an
of producers) would offer any advantage
to the bacteria in the biosynthetic

~~It~~ Whether a present
~~Whether the the~~ synthesis of
~~a given~~

We assume that

For the E_{l} whole the template H
enzyme complex is combined
with AA or free AA. ~~If this model~~

~~If the~~ enzyme ~~reactions~~ ~~~~

(The $K_{free AA}$ and K_{AA} are very large
~~these two binding~~ ~~energies~~ and K_R is
small then {this model
demands that raising the conc
of AA must suppress the formation
of the enzyme E_l} ~~provided at only~~ and
if ~~that~~ the ~~~~ of E_R is not
~~limiting~~ already saturated with
the AA ~~so that the~~ (which may
occur is the K_R content of the
cell is low.)

If on the other hand the $T - E_l$ ~~template enz~~
~~complex~~ K_{const} ~~template-enzyme complex~~ ~~free AA is very large come~~
~~small to K_R~~ the ~~enzyme~~ to K_R
~~has a~~ $K_{free AA}$ ~~of H~~ has a high
enzyme/substrate K_{const} for the precursor
of the amino acid $K_{free AA}$ and a low
const K_R for the repressor ~~Rx~~ then the
~~~~ model demands that raising the level of
~~suppress the synthesis~~ the precursor
~~must~~ enhance the production of
of the enzyme. The precursor would
then act as an inducer of the enzyme.
~~In amino acid synthesis these two~~
~~repressors furnish an alternate way~~
~~of regulations~~ the synthesis of the amino
~~acid and~~ ~~~~ the synthesis
~~of enzyme~~ would yield no ~~select~~
~~favourable~~

It is not clear why the operation
of an inducer should offer any
advantage to the bacteria and
in the biosynthetic pathway of
the synthesis of HA synthesis

It is therefore not totally therefore
in general - precursors of HA are HA
will strongly not be inducers.

We may however look for inducers
in a different system!

Gal story   the term
                             protector

Beta galactosidase   [illegible]

Inducer Kinase

The [β-] Galactosidase
story &

Paper

The model further demands
that ~~in every case~~ when ~~the~~
a bacterium that lacks an enzyme
in the biosynthetic pathway leading
for the AA ~~is~~ and is grown in the

100 to 1000 times higher rate
template can make enzyme !

presence ~~of a low conc~~   {always
~~of the AA~~ [in a chemostat ~~with~~   endnote
~~the AA as the limitation~~ in a medium
supplemented with the AA ~~which~~
~~serves as~~ (with the AA as the controlling
growth factor) ~~at its~~ the E_1 must
be formed ~~at 10 to 1000 times at te~~   at at
at least twice the rate that is ~~done~~ normal rate (relative
to the other enzymes) ~~and~~ but this likely to
be ~~found at a much higher rate~~ ✗ for sure
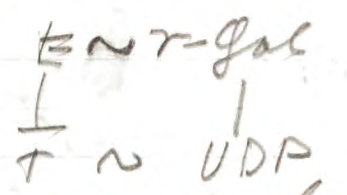~~if AA - R is kept low   most~~
~~the template must make it~~
~~at the full a two very high rate~~

[Think! If AA-R is ~~half~~ growth
rate is half AA-R must be less ✓
than 1/2 its normal value / thus enzyme
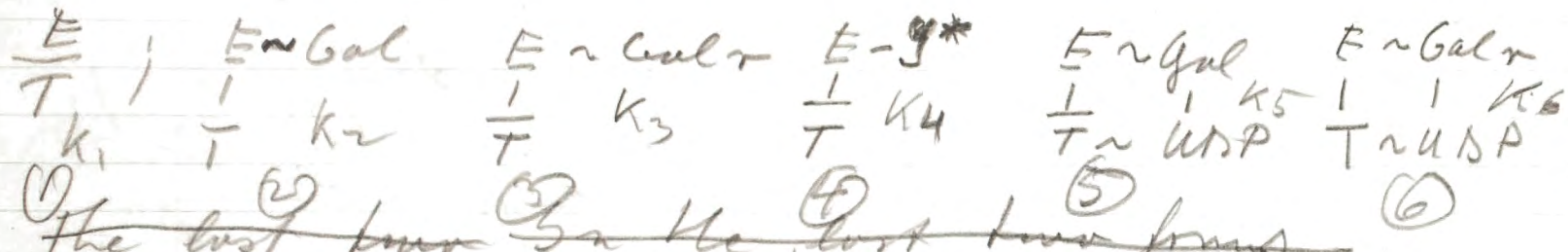$$\frac{1}{1000} = \frac{1}{1 + [\frac{AA-R}{k}]} \sim 1000$$ must double
good]

And that its [re?]formation of the enzyme
is repressed by the [reversible?]
formation of

$$E \sim r - gal$$
$$| \quad |$$
$$T \sim UDP$$

In the presence of an chemical analog g^*
of UDPGal as UDPGalr we have the
Enzyme template complex in these forms

$$\frac{E}{T} k_1 ; \quad \frac{E \sim Gal}{T} k_2 \quad \frac{E \sim Galr}{T} k_3 \quad \frac{E - g^*}{T} k_4 \quad \frac{E \sim Gal}{T \sim UDP} k_5 \quad \frac{E \sim Galr}{T \sim UDP} k_6$$

① ② ③ ④ ⑤ ⑥

The last four In the last two froms
While the enzyme - template complex
is covered as shown in 5 or 6 there is no
enzyme formation, a chemical
analog of ~~UDP Gal~~ UDP gal for which k4
is small is an Inducer because it
covers the E-T complex. UDPGAL is
both a substrate and a repressor
of the enzyme. ~~Glucose which If~~
cells are grown in glucose the UDPGal
~~tend rises and the~~ ~~say enzyme is~~
~~repressed~~ (and the UDPGalr) levels
rise and the enzyme is repressed.
~~XXXXX the "inducer may act trner~~
~~ you ≠ MG must act in most thays~~
~~that In this uncomplicated system~~
un inducer like ≠IMG might

I paper

the introduction of the enzyme
~~for galactose~~

Manual

the ~~enzyme~~ ^{lost of}~~the (bears) to the one looked~~
~~introduction of the operation of~~
~~this system was opened by that~~
~~through which we had to pass~~
~~before we could begin to get, and~~
~~two tape~~ ~~.~~ Recently Norbert
Wiener ~~opened the door to~~
~~showed the way to the unraveling~~
of this system by developing
the notion. that this enzyme
~~is not lives in~~ i$\S$^{(carried to bacteria actively}~~contains—~~^{of)}
bacteria [which contains of] not
~~is~~ because it is useful in enabling
them to exist ~~before~~ but ~~because~~
rather that this enzyme ~~is~~ lives in
the normal way ~~by photosynthesis path~~
~~may needed~~ by transf galactose
residues to the cell wall. The suggested
in ~~particular~~ that the enzyme acts
on ~~UDP Gal~~ Gal-1-P or more likely UDP Gal
I shall ~~there~~ here assume ~~for the~~
~~Gal~~ ~~without argument~~ rightly or wrongly
mainly for the sake of argument that
~~galactosidase~~ catalyzise ~~UDP~~^{is reaction}
~~which leads~~ from UDP ~~Gal~~ E ~~Gal~~ to a ~~galactose~~ ^{analog}

$$UDPGal \xrightarrow{E} (Gal UDP) ^{coupled}_{with UDP}$$

Paper

the From It follows from
and his Weiss's concept that
β⁻ Galactosida is a normal enzyme
acting on a normally present
metabolite that in the wild
type there is always present
a basal level of enzyme below
which the enzyme can not be pushed
by raising the UDP Gal conc
as long as the bacteria grow at
the normal rate, that is of course
the level at which the enzyme is fully
saturated even when fully saturated
can make only as much UDP Gal-x as
the cell needs for synthesis
of the cell wall ? but that about escape

this argument
is inclusive only if cell wall synthesis
via Gal-1 PO4 is not too important.

TPG which inhibits the synthesis
of the enzyme may act by inhibiting
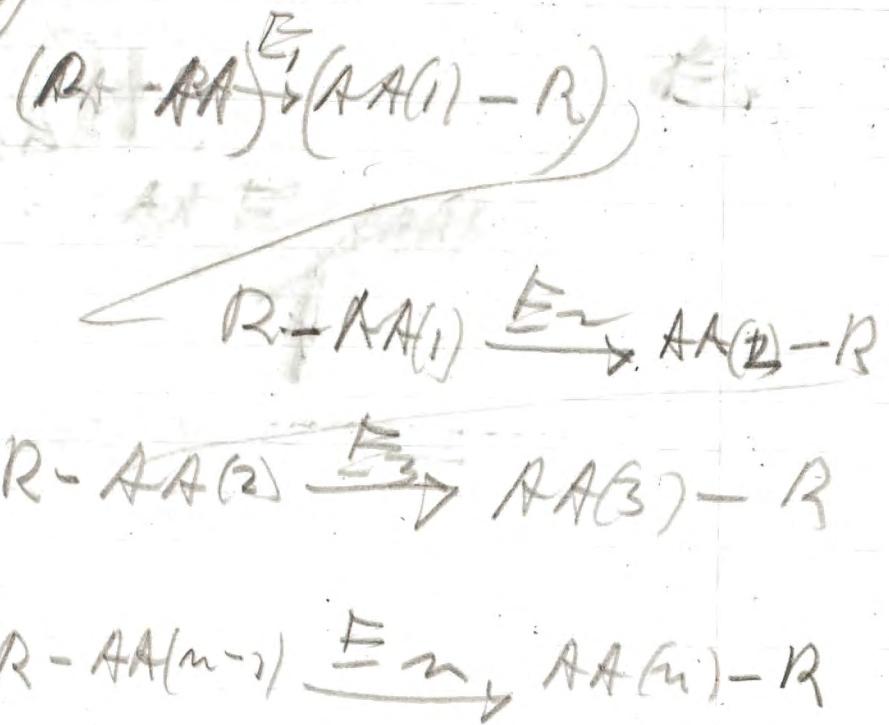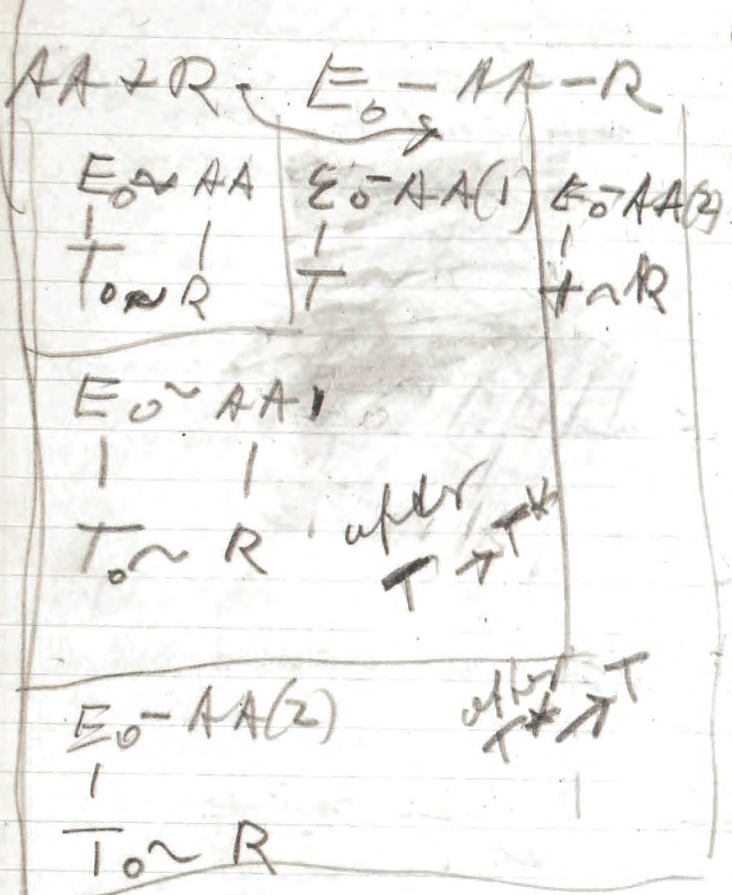E₃ and thereby slowing the
of UDP Gal to Gal-1-PO4.
conversion from

The fastest way of
checking these minimies would
be to study the effect of TPG and
+PG on the end of cell
growing on Galactose.
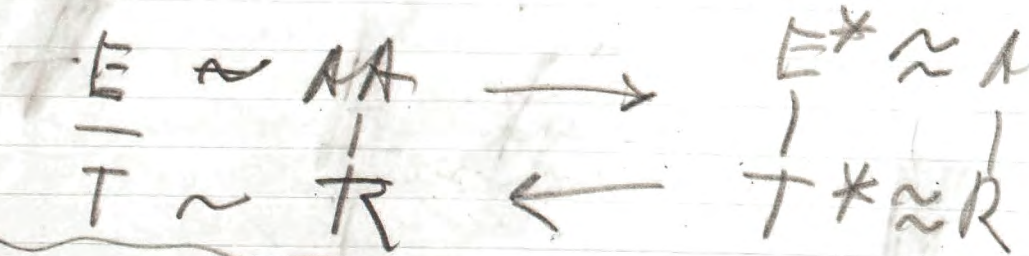
act in three different
 way ways. To see the for
one of can combine can interact
enzyme itself and thus reduces
slows the production of ΔP Gal r from UDPGal
which is a suppression of the hypothesized
But it can also combine involves with E_2 and
thus slow the production of UDPGal
Since it appears that cells (Lac⁻ mutants)
that can not make β-galactosidase
grow quite well an mannose or
inepate and therefore must be able
to build their their cell wall from Gal-1-Phy,
UDPGal can escape via a E_3 which
makes the introduction from the pool
of E_2 by $AP6$
more effective in lowering the
UDP Gal level than it would
otherwise be. It is inherent
is in this theory that the rate
of enzyme production can not increase
faster than linearly with increasing, in some,
if the inducer is acts only on one
in one spot. If it acts on two
or three spots it may rise with
the second or third power of the
the At least [in a hundred
region of concentrations only]
plus the A notations sketched above
permits us to explain the
fact that the that the
enzyme curve rises with
the second or third power at
the Mannose conc in the cell

Jumpshy ~~Not proper~~

$AA + R \rightleftharpoons E_0 - AA - R$ $\quad (R + AA) \xrightarrow{E} (AA(1) - R)$, E_1

$E_0 \sim AA \quad | E_0 - AA(1) | E_0 - AA(2)$
$\quad | \quad \quad \quad | \quad \quad \quad |$
$T_0 \sim R \quad \quad T \quad \quad \quad + \sim R$

$E_0 \sim AA 1$
$\quad | \quad \quad \quad |$
$T_0 \sim R$ after $T \rightarrow T^*$

$R - AA(1) \xrightarrow{E_2} AA(2) - R$

$R - AA(2) \xrightarrow{E_3} AA(3) - R$

$E_0 - AA(2) \qquad$
$\quad |$
$T_0 \sim R$

$R - AA(n-1) \xrightarrow{E_n} AA(n) - R$

similar for T, E where template-
- enzyme can combine with
$AA(1) - R$, $AA(1)$, $(AA2 - R)$, $(AA(2)$ $\quad AA(3)$
and $AA(3)$ etc. $\qquad \qquad -R$

and in general for $(T_n - E_n$ complex
The further we go away

$\underline{E} \sim AA \longrightarrow E^* \approx AA$
$\quad | \qquad \qquad \qquad |$
$T \sim R \longleftarrow T^* \approx R$

$AA \xrightarrow{E} AA \quad AA + R \xrightarrow{E_0} \quad$ etc

cells which contain a
high level of the enzyme
& even when the genes are lost
or mutate in [the] absence
absence of [g] inducer are
called "constitutive". ~~This
is a misnomer~~ from the
point of view of the workers
there presented this is a
misnomer. — In the wild
type the production rate of this enzyme
is represented by ~~perhaps~~ about $\frac{1}{1000}$ of its
full value, as the
result of this the wild type produces
|| deferis which cross reacts immuno-
logically ~~with the any~~ antibody is
prepared against the enzyme Just
as the ~~xxxx~~ mutants described
by Janowsky whose level of
tryptophane synthetase is represented
while only a run (and not fully restored
by the suppressor mutation)
(explain in Janowsky case
that several ±±-s are involved
and because a chem analysis
restores only of

I [PAPER]

E

AA-Adenyl

AA-adenyl

adenyl AA   E   debris

aRn-lAA   E   debris   debris

Head   tail

Unbinding

E_2 ⊙ E-tln
           T-R

Theophys

Explain here ... words ... T ...

and ... specificity ×

I Paper ~~Bi-galactosidase story~~

confirmed ?

~~Mutations can be~~

the so called constitutive strains
~~contain different forms~~ of the wild type
template T. In the mutant
the "equilibrium constant" K_1^* and K_2^*
the for the formation of the
reversible complex between the enzyme
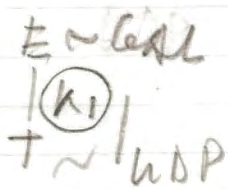~~templated~~ complex ~~is greater and~~
UDP Gal_r ~~is high~~ as well as UDP Gal
might be much higher than the
constant K for wild type

wild type   ~~constitutive~~ mutant

$E \sim Gal$ $E \sim Gal_r$ $E^* \sim Gal$ $E^* \sim Gal_r$
$|(K_1)|$ $|(K_2)|$ $|(K_1^*)|$ $|(K_2^*)|$
$T \sim UDP$ $T \sim UDP_r$ $T^* \sim UDP$ $T^* \sim UDP$

$$K_1 \text{ and } K_2 << K_1^* \text{ and } K_2^*$$

As the result a large amount of
enzyme is formed and UDPGal_r
~~might~~ reaches a high level.

~~...~~

I Paper   End of galactosidase
                                              story

Synopsis

~~suppose~~ mutants requiring
tryptophane because they
lack tryptophane synthetase
produce CRM (KRM?) (left
mound Trp) antigenic; cross
reacting

Outline

Tobbe: A theory of enzyme induction, enzyme repression and ~~gene interactions~~ repressor mutations.

Pages 18, 19, 20 ↔

Challenged by this postulate I took another look at a number of disconnected facts relating to enzyme induction to see if it were possible to bring some order in to chaos

As we shall see later it is quite understandable that in general inducers do not operate in the biosynthetic pathway that leads to the formation of an amino acid and also that they do operate in ~~other~~ certain other biosynthetic pathways, and moreover ~~above all~~ it is possible to make it at least plausible only bacteria are able to ~~carry out~~ metabolize compounds ~~biosynthetic steps~~ — as is the case in the oxidative degradation of a large number of aromatic compounds — ~~and only the enzymes involved are~~ inducers which resemble only very remotely the essential metabolites, and only they induce the enzymes which are involved in the process.

pages 57, 58, 59, 60, 61, 62, 63, 64

Galactose story

Model: (page 71)

E   R—AA
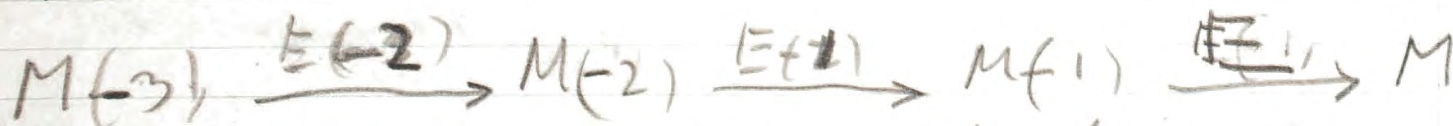
E   E~R.M E~A A   F
—   ↑ | ↑
T   T ^R   T ~ R

Summary

shall give, the various which
in my opinion must be adopted
in order to be able to understand
the formation of the a great variety of
in many frequency and the occurrence of and behaviour
of suppressor mutations and in
microorganisms and the formation of
antibodies in animals.

There might be in microorganisms
perhaps 10 and different templates
and alongside of each
there is synthetized an "enzyme"
of different specificity
We shall single out one of these
Templates T which forms an enzyme
$\underline{E}$, this enzyme forms a metabolite
M from a precursor one step removed
which we shall designate $M(-1)$

$$M(-1) \xrightarrow{\underline{E}} M$$

the metabolic pathway
which leads to M may be written as
follows

$$M(-3) \xrightarrow{\underline{E}(-2)} M(-2) \xrightarrow{\underline{E}+1} M(-1) \xrightarrow{\underline{E}-1} M$$

M might be any
metabolite which is for an
essential building block such as an
amino acid or a coenzyme or a

for the metabolite. Summary however

$$R + M \xrightarrow{E_0} \{M - R\}$$

the enzyme-template complex can reversibly combine with the metabolite M or ~~of~~ certain chemical analogues M* ~~forming which the example present in the bacteria or~~

~~different~~ ~~chemical analog~~ [The metabolite

M ~~itself~~ ~~which shall treat as its own~~ ~~chemical and~~
added to the experimenter to the ~~system~~ supernatant

Similarly the enzyme template complex can reversibly with the carrier-coupled metabolite or ~~the~~ ^{a} carrier coupled chemical analogues of the metabolite

Explain M* $(M^* - R)$

~~thus~~ ~~why we have in~~ ~~getting~~
~~to~~ In equilibrium t' we to shall have

| $E \approx M$ | $E \approx M(1)$ | $E \to M^*$ | $E \sim M(1)$ |
| 4 | $+ (K_M)$ | $+ K_{M(1)}$ | $+ \dfrac{K_{M^*}}{\sim R}$ | $\dfrac{1}{1} \dfrac{K_{M^*}}{\sim R}$ |

$$\xrightarrow{\quad E \sim M^{**} \quad}$$

forms E at $\dfrac{1}{1}$ does not form
maximal   enzyme
rate   virtually free   covered

a carbohydrate residue

In case of the amino acids as glycine
or tryptophan for instance the
precursors are known and the enzymes
the enzymes have been isolated

In the case of ornithine for instance
we have $M(-1) =$ citrulline; $M(-2) =$
ornithine; $M(-3) =$ acetyl ornithine.

According to the notation here
prescribed the enzyme - Template
complex for which we shall write

$$\frac{E}{T} \quad ; \quad \frac{E \sim M}{T \sim R}$$

if it were not not repressed in
the manner that will be presently
explained would turn (in biochemistry) enzyme
at a rate of about 10000 higher
than do the templates
in growing bacteria
actually do.

Normally however this comes about
in the following manner the metabolite M
is coupled to a radical R [conceivably
might be when $M =$ galactose and conceivably
adenyl when M is an amino acid]
by a coupling enzyme which is specific

$$E_0$$

as well as in other microorganisms

... by a factor of about 10^4.

However if this ~~the~~ is the only effect that an inducer M^{**} has then ~~act may be shown~~ ~~though enzyme formation~~ as it may be shown the rate of enzyme prod is a function of the inducer concentration can not rise faster than linearly.

~~And now we~~ A chemical analog M^{**} of the metabolite can however raise the enzyme ~~leve~~ not only by protecting the enzyme-template complex but also by inhibiting ~~an~~ enzymes which makes an inhibitor

$$M + R \xrightarrow[]{E_0} \{M-R\} \xrightarrow{E_1} \{M(1)-R\}$$

$$\uparrow M^* \qquad\qquad \uparrow M^*$$

(1)

It is likely that during evolution mutations that occurred in template & which sometime ~~decreased~~ increased some time decreased

We shall designate the repressor
{R-M} with rep. and (R-M*) with rep(*)
And accordingly designate
the acquilibrium constants for
binding with the E-t complex
with K_M, K_{M*}, K_{M**}, K_{rep}, K_{rep*} (6)
By deriving the relations for the
equilibrium concentrations
for the virtually free and the
covered forms of the E-t complex
one can see that if K_{M**} is small
compared to both K_{rep} and K_{rep*}
then at comparably small concentra-
tion of M** almost all the E-t
complex might be present
in the virtually free form and
enzyme production will be
greatly enhanced.
M* may then be called an inducer
of the enzyme. Since in many cases
enzyme production is repressed to
about $\frac{1}{10^4}$ of the enzyme
production of the free enzyme-template
complex an inducer may
enhance enzyme production

What are superior antibodies
for dog

Tryptophane synthetase

P_2 Strander case

Beta galactosidase   cust
antibody. ⑪

$$\frac{E}{T} \quad \frac{E \sim g}{T} \quad \frac{E \sim galactose}{T} \quad \frac{R}$$

$$(H_m - R) \xrightarrow{E_m} (H_m - R)$$

$$[H_m - R] = r_m \qquad [E_m] = e_m \qquad e = [E_m]$$

$$\frac{dr_m}{dt} = const \cdot e_m - \frac{r_m}{\tau_m}$$

$$\frac{de}{dt} = const \cdot N_m$$

$$\{ H - Prot \} = Antigen$$

$$E \approx H - Prot \qquad \frac{E}{T} \approx \frac{H}{|}$$

$$\frac{T}{\sim} Prot$$

$$\frac{A}{T} \qquad \frac{A \sim H_m}{T}$$

$E \alpha$ galactose

$\frac{E}{T}$ $\frac{E \sim g}{T}$

$\frac{E}{T}$ cons

$T \Delta B$
must produce
Enzyme Template
complex

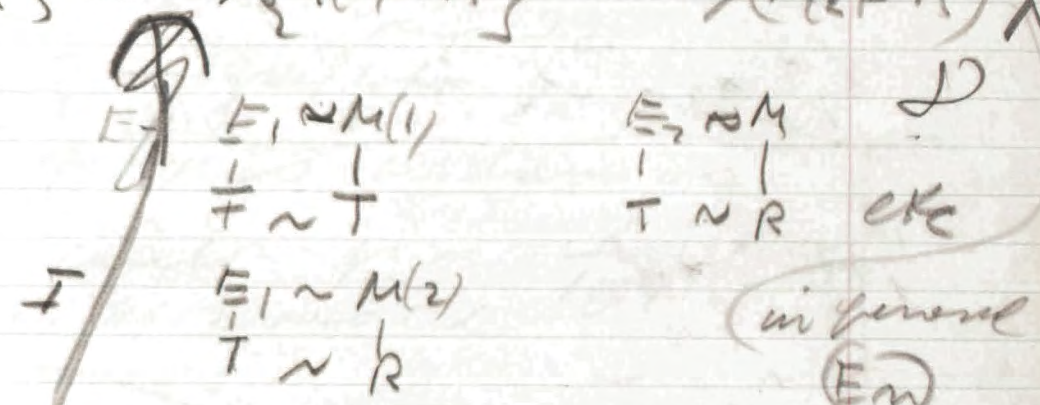in embryo and
newborn only

$$\frac{A}{T} \approx \frac{H}{|} \approx Prot$$

Secondary response

Chase

Curios phenomenon

the value of the equilibrium
constant K_{rep} ~~were each time~~
it were suppressed by mutations
occurring in other loci and
that this had to a series of
enzymes and ^new^ chemical analogs
of the metabolite m so that
totally on we have in here a
~~chains~~ metabolic chains of
this sort M ~~M~~ $\xrightarrow{E}$ $M(1)$ $M(1) \xrightarrow{E_2} M(2)$

$M + R \xrightarrow{E_0} \{M - R\} \xrightarrow{E(1)} \{M(1) - R\} \xrightarrow{E_2} (M(2) - R)$

$\begin{array}{l} \underline{E} \sim M(1) \\ \overline{T} \\ T - \end{array}$ $\begin{array}{l} E_0 \sim M \\ \overline{T} \sim R \\ E_0 \sim M(1) \end{array}$

$\begin{array}{l} \underline{E} \sim M(2) \\ \overline{T} \end{array}$ $\begin{array}{l} E_0 \sim M(1) \\ \overline{T} \sim R \end{array}$

E_T $\begin{array}{l} E_1 \sim M(1) \\ \overline{T} \sim T \end{array}$ $\begin{array}{l} E_2 \sim M \\ \overline{T} \sim R \end{array}$ etc

$?!!$ $E_0 -$ $\overline{I}$ $\begin{array}{l} E_1 \sim M(2) \\ \overline{T} \sim R \end{array}$ (in general

$\underline{E_n}$

$M \xrightarrow{E} M(1)$ $M(1) \xrightarrow{E_2} M(2)$

$M^* \xrightarrow{E} M(1)$ $M^*(1) \xrightarrow{E} M^*(2)$

$\xrightarrow{II} \boxed{\dfrac{E}{T}(1) - M^*}$

branched chains
for removed remittance

$\widehat{E_n}$

Enzyme Kinetics
for Autocatalysis

Saturation $\dfrac{1}{1+\frac{r}{R}}$

if enzyme is saturated! $r \propto E_0$

① $\dfrac{dE}{dr} = A\,\dfrac{1}{1+\frac{r}{R}} - \dfrac{E}{\tau}$

$r = const\, E$

$\dfrac{E}{\tau} = \dfrac{A}{1+\frac{r}{R}}$ $\qquad \left(1+\dfrac{r}{R}\right)\dfrac{E}{\tau} = A$

$1 + \dfrac{const\, E}{R}\,\dfrac{E}{\tau} = A \qquad E = \sqrt{A}$

$\Delta F = RT \log K$

Arguments for exceedingly small value

N/cc 10^{-16} mol $\times$ v

10^{-6} gm/cc $-16 \div 10^{15}$

10^{-8} mol/cc $10^{-6} \div 10^{-2}$ $\sim 10^{13} e^{-\frac{\Delta H}{RT}}$

10^{-8}

10^{4}

10^{-12}

$-12 \times 2.3 \approx 24$

Concentrations about micrograms/cc

$\gamma gm/cc$

or 1 gm/liter

Concentrations as low as
1 μ/cc still could give $\frac{1}{10^4}$ free
at binding ene of 12 000 cal

Replication means all templates F–B
within a cell must be destroyed

Assume 10 templates per cell

Orders of Magnitude

When we start with very low
M "free decreases" and later (∝ M
saturates" + free fraction

$$\frac{1}{free} = \frac{1 + \frac{MM}{Km} + \frac{C \times M}{Krep}}{1 + \frac{M*}{K(M*)}}$$

half saturation

for $1 = \frac{M'}{K(M)}$

Orders of magnitude
not taken
Inducer M*

rate of enzyme
prod doubles where
M is reduced from
very high conc to

$$V = \frac{1 + \frac{M*}{K(M*)}}{1 + \frac{M*}{K(M*)} + \frac{rep}{Krep}}$$

V

$$\frac{M*}{K M*} \ll \frac{rep}{Krep}$$

2/3

when $\frac{M*}{K(M*)} = \frac{rep}{Krep}$

Constant for phenylethyl gal
10^{-5} molar

$M = 100$ gm 10^{-10} molar or more
10^{-8} gm/liter

Oxymoglobin diagram

$$\text{free } O_2^* = \frac{1}{1 + \frac{M^*}{K_1} + \frac{rep}{K_{rep}}}$$

virtually free

① $O_2^* \neq [Me] = \dfrac{1 + \dfrac{M^*}{K M^*}}{1 + \dfrac{M^*}{K_M^*} + \dfrac{rep}{K_{rep}}} = |v_1 v_2|$

② $\dfrac{M^*}{K(M^*)} \ll \dfrac{rep}{K_{rep}}$ and $1 \ll \dfrac{rep}{K_{rep}}$

($M = M^*$)

$3 \quad \dfrac{M^*}{K(M^*)} = M$

If we raise M and E_0 for transat,

If we raise M^* (and E_0 for trans

saturation just $1 \ll \dfrac{M^*}{K M^*} \ll \dfrac{rep}{K_{rep}}$

so that

rep rises rep to $\dfrac{M^*}{KM^*}$

$\dfrac{M^*}{KM^*} > \dfrac{M^*}{KM^*}$

$M = M^*$

4) $rep = const \; M(on)$

We assume E_0 not near saturation

then $\dfrac{rep}{KM^*} = const \; M^*$ and $\dfrac{M}{K(M)} \ll \dfrac{rep}{K_{rep}}$

(1) $1 \ll \dfrac{M^*}{K(M^*)}$ then "free" indep of M^*

(2) $1 \gg \dfrac{M}{K(M)}$

5) $\dfrac{[M]}{K(M)} = 1$

6) $\dfrac{M}{K(M)} \gg 1$ free falls with increasing M rises with

another 7) $\dfrac{M}{K M} \ll 1$ free must thus always become very large if we make M low enough

$$V = \dfrac{1 + \dfrac{KM}{KM}}{1 + \dfrac{M}{K} + \dfrac{rep}{K_{rep}}}$$

20% of dry weight

enzyme in leaf

which has 15,000 enzyme

1000 ont. dens. 700 g dry weight

15,000 unit enzyme in 700 g

1 mg Z   1.6 10⁵ units   of dry weight

40 % of total protein —

$$\cancel{K_{rep} = 10^{-9} M}$$

$$\frac{rep}{K_{rep}} = 10^4$$

$$\cancel{rep = 10^{-4} M}$$

this means that 10^{-5} M/ber
producer may double bord
level

$\text{for } M \cancel{6} \text{ if was } \cancel{10} / \text{ nolor}$
10^{-2} would double 1 av
ontside cure

and 10^{-2} molar would raise
by a factor 1000

(What is rep conc. ?)

Under acid etc, these with
can not represed by
raising "Beginner" cure
$M \sim K_M$ or $K_{M} = 10^{-5}$

$M = 10^{-5}$ or Mol weight 100
10^{-3} gm per liber
may be 10 g or 10^{-6} gm
outside 100 × 10 g / mile
where M is lower increasing any—
will depress enzyme

Autocatalyst new May 8/53

~~Translate ... enzyme~~ (Autocatalyst)

number of templates 1

$$\frac{dE}{dt} = \frac{\alpha K E}{1 + \frac{r_0}{K}} - \frac{E}{\tau_E}$$

$$\frac{dr}{dt} = \beta E - \frac{r}{\tau_{rep}} \qquad q \frac{\alpha K}{r} = \frac{E}{\tau_E} \quad \boxed{0 < q < 1}$$

$$\boxed{r = \beta E \tau_{rep}} \qquad \frac{\alpha K}{\tau}$$

$$q \alpha K \tau(E) = \beta E^2 \tau_{rep} \qquad q \alpha K \tau(E) = \frac{r}{E}$$

50 fold increase

$$10^{13} \left(e^{-\frac{Q}{RT}} = 10^{-5} \right)$$
$$e^{-\frac{Q}{RT}} \qquad 10^{-18}$$
$$\frac{Q}{RT} = \qquad \begin{matrix} 18 \times 2 \\ 40\ 600 \\ 4\ 000 \end{matrix}$$

Rep E →

10^{-5} Mol/liter   Bulärs

$\emptyset - TS$ $e^{-F_1} e^{-1.0}$

ϕ

e^{-3}

$R T \times 7$

$2\pi pw$ $1000 x$

$N = \dfrac{m}{u}$

$1u = 10^{-3} e$

$N = \dfrac{n}{10^{3}}$

$\dfrac{N}{cc} \approx 10^{+4} cm \; 10^{-15} = 10^{12} e^{-10} (1=f) \tfrac{1}{2}$

$= 10^{19} \, 510^{-5}$

$\boxed{N = 510^{19} f \, f \tfrac{1}{2}}$

$f = \dfrac{1}{1 + \dfrac{N/u}{K}}$

$N =$

510^{8}

$f = \dfrac{1}{1 + \dfrac{A(1-f)}{K}}$

$\dfrac{510^{22}}{2}$ $\boxed{\dfrac{1}{1000}}$

$f \sim \dfrac{1}{1 + \dfrac{A}{K}}$

$8 \cdot 10^{23} \, 10^{-5} \, 10^{-3} \, 10^{4} \, 10^{-15} = \tfrac{1}{12} 10^{12} e^{-\frac{\Delta H}{RT}}$

$10^{27} \, 10^{-23} \, 10^{-14} = e^{-\frac{\Delta H}{RT} 10^{12}}$

$\underline{10^{7}}$

$$K_D = 10^{-6}$$

$$K_{(12)} = 10^{-5}$$

$$K_{1, \text{\#\#}}$$

$$K_{\circled{12}} = 10^{-11}$$

---

temper $\frac{1}{10}$ $\boxed{\text{per liter}}$ $M = 100$

$10^{-4}\,\text{gm}$ $K = 10^{-6}$ $\Delta H_1 = \Delta H_2 =$

$\qquad\qquad M$ $\qquad\qquad 12\,\text{a.s.s.}\,\text{cal}$

$$\frac{K_{rep}}{K_M} = 10^{-5}$$

$$\frac{1 + \dfrac{M}{K_M}}{1 + \dfrac{[rep]}{K_{rep}}}$$

$$\frac{[rep]}{K_{rep}} \approx 10^4$$

$\boxed{10^{-6}\,\text{mol} = 0.1\,\text{mgm}}$
$100\,\text{gm an weight}$ $\boxed{100\,\text{g}}$

$\sout{\text{under}}$ $\boxed{M = K_M}$ $[rep] \gtrless [K_M\ K_D}$

$$\frac{[rep]}{10^{-5}\,K_M} = 10^4 \qquad \boxed{[rep] = 10^{-1}\,K_M}$$

$$\text{Molass} \qquad A \qquad B \qquad A\text{-}B$$

$$K_1 \quad \nu \, 10^{-16} \rightarrow \tfrac{1}{2} 10^{13} e^{-\frac{\Delta H_1}{RT}}$$

$$N_{12} \quad \nu \, 10^{-16} \rightarrow \tfrac{1}{2} 10^{13} e^{-\frac{\Delta H_1 + (\Delta H_1 + \Delta H_2)}{RT}} \qquad \tfrac{1}{q} = 10000$$

$$\frac{K_{12}}{K_1} = \frac{1}{q} e^{-\frac{\Delta H_2}{RT}}$$

$$\boxed{\frac{K_{12}}{K_1}} = 10000 e^{-\frac{\Delta H_2}{RT}} = 10^4 e^{-20} \boxed{10^{-5}}$$

$$\boxed{K_1 = 10^{-6} \, \text{mol}} \qquad 10^4$$

$$N_1 \quad \nu \, 10^{-16} = \tfrac{1}{2} 10^{13} \cdot 10^{-20 \cdot 1.5}$$

$$N_1 = \tfrac{1}{2} \qquad \tfrac{1}{2} 10^{-4} 10^{16} 10^{13} e^{-20}$$

$$\text{mol/liter} = \tfrac{1}{2} 10^{25} e^{-20}$$

$$K = \frac{10^3}{6 \cdot 10^{23}} \tfrac{1}{2} 10^{25} e^{-2\%}$$

$$\boxed{10^{-6}} = K_1 \qquad \text{(1)} \qquad 12,000$$

$$+ = \frac{10^9}{10} \cdot 10^2 = 10^{20} - e^{-23}$$

$$\| = \qquad 10^9 e^{x} \qquad e^{\frac{12000}{6000}}$$

$$x$$

$$9.23 \qquad \text{mol} \left(10^{12} \right)$$

Antihelixes (general theory)

$\frac{dE}{dt} = bq \sqrt{1 + \frac{r}{K(M)}} \bigg/ 1 + \frac{r\epsilon p}{} - E$

$\frac{E_0}{q} = bq \frac{\epsilon}{10^4 (m+1)}$

10 days

$\frac{dr}{dt} = a - \frac{1}{1 + \frac{r}{K(M)}} E - \frac{r}{\rho}$

$K \epsilon p$

$1 + \frac{r_0}{K(M)}$ N

for $t = 0$ $E = E_0$ $r = r_0$

$1 + \frac{r_0}{K(M)}$ N m

$r_0 = K(M)$

at $t = 0$ $\dfrac{2}{1 + \frac{r\epsilon p}{K r \epsilon p}} = \frac{E_0}{E}$

$\frac{q}{\epsilon} 10^4 = E_0$

$\boxed{\epsilon = 10 \text{ days}}$

$t = 0$ $a = \frac{bq \epsilon \cdot 10^4}{2} + \frac{K(M)}{\rho}$

$0 = a - \frac{E_0}{m} - \frac{r_0}{\rho} \left((m-1) K(M) \right)$

$0 = a - bq\epsilon \frac{1}{104} - \frac{r_0}{\rho}$

$K(M) = 10^{-6}$

$a = bq\epsilon$

$\frac{r}{r_0} = \frac{a\rho}{(m-1) K(M)} = \frac{r_0 + bq\epsilon \frac{1}{104}}{(m-1) K(M)}$

$= 1 + bq\epsilon \frac{1}{10^4} (m+1)$

Antibubbles (good theory)

$\frac{dE}{dt} = bq \sqrt{\dfrac{1 + \frac{r}{K(M)}}{1 + \frac{r}{veb}}} - \frac{E}{\mathcal{E}}$

$\boxed{\dfrac{E_0}{q} = bq \dfrac{\mathcal{E}}{10^4 (m+1)}}$

$\frac{dr}{dt} = a - \dfrac{1}{1 + \frac{r}{K(M)}} E - \frac{r}{\mathcal{E}}$

for $t = 0$ $E = E_0$ $r = r_0$

$\left(1 + \dfrac{r_0}{K(M)}\right) N$

$r_0 = K(M)$

at $t = 0$ $\dfrac{2}{1 + \frac{r_0 b}{K_{rep}}} = \dfrac{E_0}{\mathcal{E}}$

$q/\mathcal{E} \, 10^4 = E_0$

$\boxed{\mathcal{E} = 10 \text{ days}}$

$t = 0$ $a = \dfrac{bq \mathcal{E} 10^4}{2} + \dfrac{K(M)}{\mathcal{S}}$

$0 = a - \dfrac{E_0}{m} - \dfrac{r_0}{\mathcal{S}} \left((m-1) K(M) \right)$

$0 = a - bq \mathcal{E} \dfrac{1}{10^4} - \dfrac{r_0}{\mathcal{S}}$

$K(M) = 10^{-6}$

$a = bq \mathcal{E}$

$\dfrac{r}{r_0} = \dfrac{a \mathcal{S}}{(m-1) K(M)} = \dfrac{r_0 + bq \mathcal{E} \frac{1}{10^4}}{(m-1) K(M)}$

$= 1 + \dfrac{bq \mathcal{E} 10}{(m+1)}$

Faraday Talmadge

$$10^{12} e^{-\frac{4H}{RT}} = 10^{-5}$$

$$e^{-x} = 10^{-17}$$

$$e^{-\frac{40}{RT}}$$

Does it make a difference when we elute?

Experiment:

Antibody response when antigen removed at a time "t" rabbit makes inactive antibody.

Experiment

is all the additional of glob antibody 2, 2 1)

Ampleurdy pound flow after
long.   only little Δy gets into
nucleus; rather was embraces photoserals
in a few drugs and others annihie.

Δp inside

The Representation
Acquilibrain;
Sites avalinded with rep

$$\frac{\dfrac{Rep\,n.p}{K_{tene}}}{1 + \dfrac{Rep\,n.p}{K_{tene}}} = Rep$$

$$\left. \frac{rep}{K} \cdot \frac{rep\,low}{1 + \dfrac{rep\,low}{K}} = 1/2 \right\}$$

$$\frac{rep\,(low)}{K} = \frac{1}{2} + \frac{1}{2}\,\frac{rep\,low}{K}$$

$$\frac{rep\,low}{K} = 1 \quad \left\| \quad \frac{rep}{K} = \quad \;\; s \perp \quad s\,\frac{rep}{K} \right.$$

$$(1-s)\,\frac{rep}{K} = s$$

$$rep = \frac{s}{1-s} \qquad s \approx 1$$

$$\Delta \geq \int^{n.p} g$$

rep,n.p [g hold]

$$= \frac{1}{1-s}$$

Antibody formation check

$$\frac{dE}{dt} = bq \frac{\left(1 + \frac{r}{K(M)}\right)^{-m+1}}{1 + \frac{r_{rep}}{K_{rep}}} \rightarrow \frac{E}{\varepsilon} \quad 1.)$$

$$\frac{dr}{dt} = a - \frac{\varepsilon}{1 + \frac{r}{K(M)}} E - \frac{r}{\rho}$$

$$\frac{r_o}{K(M)} = M \qquad \cancel{E_o = \left(a - \frac{r_o}{\rho}\right)(m=1)}$$

$$r$$

$$1.) \quad \frac{E_o}{\cancel{q}} = bq\varepsilon 10^{-4}(m+1)$$

$$r_\infty = a\rho \qquad r_o \parallel \cancel{a\rho}\left(a - \frac{1}{1+m}E_o\right)\rho = r_o$$

$$r_\infty = a\rho = \left(\cancel{am + q}\right)K_m + \frac{aq}{1+m}\varepsilon 10^{-4}\rho$$

$$\boxed{\frac{r_\infty}{r_o} = 1 + bcq \,\varepsilon\rho}$$

we are all right !! $\longrightarrow$ if $\rho_{rep} \ll \rho_M$ then ρ and ε are elimination constants.

✳ 100 enzymes

1000   10 molecules or
2×10^{-13} gm Protein   100,000 mol weight

$2 \cdot 10^{-18}$ mols

$6 \cdot 10^{23} \cdot 10^{-18} = 6 \cdot 10^{5}$ $\boxed{10^{6}}$

{ 1000 mg   1000 molecules
  1000 mg   10   molecules

Why is there a minimum
growth rate? ? ?   great!

10,000 enzymes/τ   1000 min
            100 enz per min

1/per sec   100,000 mol weight   1000 residues

n = 100 Arg · # $\dfrac{\tau}{n} = \boxed{1000 \text{ doublt}}$

n 10 ⟨⟩

$\tau = n \times x = 10$ sec

$10^{-12} \; 10/n \; 10^{13} e^{-X} = \dfrac{1}{n \cdot 10^{-13}}$

$e^{-X} = \dfrac{10^{-13}}{n \cdot 10}$

10^{-14} or 10^{-15}

unclear rep curve, & in
"exponential" experiments?,

should be really 10^4 on our assump
+ dans.—

Shape of curve important!

~~(As...)~~

$E(\Sigma)$ very important

$$\frac{Vector}{K} = \frac{rep}{K gene}$$

It means one by one put
on! If we get large increase in
rep when growth is slow. No!

Probably
unclear!

$M \xrightarrow{E_1} M(1)$

$(M_1) \xrightarrow{E_n} M$

$(M-1) \xrightarrow{E_n} M(n)$

will be
transferred

E_n+1
will more

$M^+(n)$

$rep(n+1)$ low

$rep(n)$ low

$rep(n-1)$ low

$rep(n-1)$ high

high

E_n+1 will
be
induced

$T(n+1) \quad free \qquad \dfrac{\parallel}{1 + \dfrac{rep(n-1)}{K(n-1)}}$

M

$M + R \xrightarrow{E_0} \{M - B\} \xrightarrow{E_1}$

into polymer

$M(1)$

is induced

if E_2

there is an easy answer isn't
gather?

$100/l$ 10^{14} molecules/cc

$10^{14} \times 10^5$ 10^{-16} $\boxed{\dfrac{1}{1000} \text{ of sec}}$

10^{1}

$\boxed{n \text{ boxes}}$ after x sec probab $\dfrac{x}{1000}$

10^{-12}

$\left(\dfrac{x}{1000}\right)$ $K_a = $ Probab to evaporate $\times \frac{1}{2}$

prob of empty e^{-n}

prob of not empty $1 - e^{-n} = 1 - e^{\frac{t}{1000}}$

$\left(1 - e^{-\frac{t}{1000}}\right)^{100} =$ $\boxed{10 \text{ sec}}$

$= \frac{1}{e}$

$\boxed{[\text{sec}]}$ better $1000/l$

of 1 out of 100 evaporates then trouble

life time $10^{12} e^{-y} = \tau$

10^{-6} molar

100 gm mol/lit 10^{-4} amp/cm $y = \lg \tau \times 12 \times 2.3$ $\boxed{24}$

100 boxes $100/l$ $\dfrac{3.6}{27.6}$

$K = 10^{-11}$ transported $\dfrac{1}{1 + \dfrac{10^{-7}}{10^{-11}}}$ $\boxed{12,000}$

10^{-4} is sec

Bence Jones Proteries (Mill?)

60% of serum of globulin (multiple
instead of 1/2 to 2% in normal/urine myeloma...)

Inject 10 gm Bence Jones excreted in
half the cases (and except 10 to 15 ...)
2 5% excrete excrete both of — Glob and
Bence Jones .

1/2 excrete Bence Jones and 1/2 Gamma glob.

$$M(-3) \longrightarrow M-2 \qquad M-1 \qquad M$$

$$M+R \xrightarrow{E} \{M-R\} = rep(0)$$

$rep(0)$

$$rep\ 0 \xrightarrow{E(1)} rep\ 1 \xrightarrow{E_2} rep\ 2$$

$$M \xrightarrow{E_1} M_1 \xrightarrow{E_2} M_2$$

$$M_1^*$$

M_1 or M_1^* is inducer of E_2

increases $rep(0)$ decreases $rep(1)$
probably induces E_1 also

Drug tolerance

Expression   pre Metabolites
                    post metabolites

$$\frac{dE}{dt} = aT(1 - p_n - q_n) \frac{1}{1 + \frac{r}{K_1}} - \frac{E}{\tau_E}$$

$$t_0 \qquad \boxed{E_0 = aT\tau_E \frac{K_1}{r_0}(1 - p_0 - q_0)}$$

$$\frac{dr}{dt} = bE - \frac{r}{\tau_r} \qquad \boxed{p_n = 0.2 \quad q_1 = 0.1}$$

$$\boxed{r_0 = bE_0\tau_r} \qquad \boxed{\begin{array}{l} E_p^2 = \frac{a}{b} \frac{\tau_E}{\tau_r} TK_1(1 - p_0 - q_0) \\[4pt] r_p^2 = ab\, T\tau_E\tau_r K_1(1 - p_0 - q_0) \end{array}}$$

$$E = E_0 + \Delta t \qquad E \equiv 0 \qquad \boxed{\frac{r^2}{K^2} = \frac{ab\,T}{K}} $$

$$\frac{dE}{dt} \equiv 0 \qquad \Bigg\}$$

$$\frac{r}{K}$$

$$\tau_E = 26 \text{ day}$$

$$\tau_n = 1 \text{ day}$$

$$\frac{r_1}{K_1} = 10^4$$

$$\frac{r_1}{K_2} = 10^2$$

$$\frac{dA}{dt} = p_2 Ta \frac{1}{1 + \frac{r}{K_2}} - \frac{A}{\tau_A}$$

$$p_2 = 0.2 + 0.7 \times 0.2$$

$$q_2 = 0.1 + 0.7 \times 0.1$$

3 day   6 day   4 day

$$\frac{d E_n}{dt} = a \frac{T}{1 + \frac{r}{k_{E_n}}} - \frac{E_n}{\tau_E}$$

$$\frac{r_0}{k_{E_n}} = 10^3$$

dumps

$$rep(n-1) \xrightarrow{E(n-1)} rep(n) \xrightarrow{E_n} \frac{rep}{(n+1)}$$

$$\frac{dr}{dt} = b E - \frac{r}{\tau} \quad \text{- 1 day}$$

$$\boxed{r_0 = b E_0 \tau_r}$$

$M(n)$

r

$$E_0 = \frac{a \sqrt{k e} \tau_E}{r_0}$$

$$\boxed{1 + \frac{r}{k_E} \approx \frac{r}{k_E}}$$

τ is same at templates?

$$E_0^2 = \frac{a}{b} \frac{\tau_E}{\tau_r} k_e T$$

$$r_0^2 = a b \tau_r \tau_e k_e T$$

$E = E_0$ than ub

E_0, r_0 at $t = 0$ $\quad || \quad$ at $t = 0 + \Delta t$, $E \equiv 0$ now r falls

$$e_1 = \frac{E_1}{\beta \tau_E} \qquad e_2 = \frac{E_2}{\beta \tau_E}$$

$$\cancel{\mathcal{L} \times \tau_x E_1(\theta)} \qquad E_1(0) = \beta \tau_E$$

$$x(0) = \alpha \beta \tau_E \tau_x$$

$$\frac{x(0)}{K} = 10^4$$

$$\frac{de_1}{dt} = \frac{1}{\tau_e} \frac{1}{1 + 10^4 e_2} - \frac{e_1}{\tau_e} \quad \Big\}$$

$$\frac{de_2}{dt} = \frac{1}{\tau_e} \frac{1}{1 + 10^4 e_1} - \frac{e_2}{\tau_e}$$

$$t=0 \quad \begin{matrix} e_1 = 1 \\ e_2 = 0 \end{matrix} \Big\} \quad \cancel{\text{////}}$$

$$\boxed{\tau_e = 1}$$

$$e_1 = e^{-\alpha t}$$

$$N e_2 = 1$$

$$N = 10^4 \quad \Big\rceil$$

$$N = 10^6 \quad \Big\}$$

$$\underline{\beta_5 =}$$

$$\frac{k\alpha}{\tau_A} = \beta_1 T \frac{1}{\frac{r_2}{r_0}\frac{r_0}{K_2}}$$

$\frac{k_2}{}$

$$\frac{dx_1}{dt} = \alpha E_1 - \frac{x}{\tau_x}$$

$$\frac{dy}{dt} = \alpha E_2 - \frac{x}{\tau_y}$$

$$\frac{dE_2}{dt} = \frac{\beta}{1 + \frac{x}{K_x}} - \frac{E_2}{\tau_E}$$

$$\frac{dE_1}{dt} = \frac{\beta}{1 + \frac{y}{K_y}} - \frac{E_1}{\tau_E}$$

τ small

$\tau_x = \tau_y$

$\tau_E >>> \tau_x$

$K_x = K_y$

$x = \alpha E_1 \tau_x$

$y = \alpha E_2 \tau_x$

$$\frac{dE_1}{dt} = \frac{\beta}{1 + \frac{\alpha \tau_x E_2}{K}} - \frac{E_1}{\tau_E}$$

$$\frac{dE_2}{dt} = \frac{\beta}{1 + \frac{\alpha \tau_x E_1}{K}} - \frac{E_2}{\tau_E}$$

$E_2 = 0$ $\qquad E_1 = \beta \tau_E$ $\qquad \dfrac{\alpha \tau_x E_1(0)}{K} = 10^4$

$t = 0$

$$N e_2 = \frac{1}{e_1} - \frac{e^{-t}}{e_1} = 1$$

$$\boxed{1 - e^{-t} = e_1}$$

$$\frac{de_1}{dt} = \frac{1}{1 + \frac{1}{e_1}\{1 - e^{-t}\}} - e_1$$

$$\boxed{e_1 = 1 - \frac{1}{N} - \sqrt{}}$$

$$\frac{e_1}{e_1 + 1 - e^{-t}} - e_1$$

$$\frac{1}{N} - e_2 e'$$

$$e_1 = 1 - \frac{1}{N}$$

$$\frac{1}{N} - e_1 = \frac{1}{N} e^{-\tau}$$

$$\frac{1}{1 + N - 1}$$

$$\frac{1}{N} - N + \frac{1}{N} = \frac{1}{N} e^{-\tau}$$

$$-y' = - \boxed{\frac{1}{N e_1} e^{-t}} \quad -y + \frac{1}{Ne}$$

$$+y = +y + \frac{1}{N e_1} \qquad \frac{1}{e_1} - N e_2 = \frac{1}{e_1} e^{-t}$$

$$\frac{1}{N} - e_2 e_1 = \frac{1}{N} e^{-t} \qquad 1 - (Ne_2)\left(1 - \frac{1}{N}\right) = \frac{N}{N-1} e^{-\tau}$$

$$1 - (Ne_2) e_1 = e^{-t} \qquad 1 - 1 + \frac{1}{N} = \frac{N}{N-1} e^{-\tau}$$

$$\boxed{1 - 1 + \frac{1}{N} = e^{-\tau}} \qquad \frac{1}{N}$$

$$\frac{1}{1+10^4 e_2} = -\alpha e^{-\alpha t} + e^{-\alpha t} = (1-\alpha)\, e^{-\alpha t}$$

$$\frac{1}{(1-\alpha)}\, e^{\alpha t} \qquad \boxed{e_1 \sim e^{\alpha t}}$$

$$\frac{1}{1-\alpha}\, e^{\alpha t} = 1 + 10^4 e_2$$

$$10^{-4}\left(\frac{1}{1-\alpha}\, e^{\alpha t} - 1\right) = e_2$$

e_2 analyses
e_2 ring theory
of operators

~~~~~~~~~~~~~~~~~~~~

$$\frac{de_1}{dt} = \frac{1}{1+Re_2} - e_1$$

$$\frac{de_2}{dt} = \frac{1}{1+Ne_1} - e_2 \;\;\cong\;\; \boxed{\frac{1}{Ne_1} - e_2}$$

$$\frac{dy}{dt} = \text{const} - y$$

$$\frac{N-P}{N}$$

$$e^{-\alpha t}\frac{1}{Ne_1} - \left\{\frac{1}{Ne_1} + \frac{1}{Ne_1} e^{-t}\right\}$$

$$-e^{-t} = -e^{-t}$$

$$e_2 \lessgtr \frac{1}{Ne_1}$$

$$c - y = \eta$$
$$\eta = c - y$$
$$-\frac{d\eta}{dt} = \eta$$
$$\eta = \eta_0 e^{-t}$$

$$\boxed{\frac{d}{dt} e_2 = \frac{1}{Ne_1} - \frac{1e^{-t}}{Ne_1}}$$

$$\frac{1}{Ne_1} - y = \text{const}\, e^{-t} \qquad \frac{1}{Ne_1} - y = \frac{1}{Ne_1} e^{-t}$$

$$1 - \frac{N}{E} e_1 = e^{-t}$$

$$\tau = \log \frac{1}{1 - N/E}$$

$$\tau = \log\left(1 + \frac{N}{E}\right) = \frac{N}{E}$$

$$t_1 = \frac{N}{2E}$$

$$E \gg N$$

$$1 - \frac{N}{E} e_1 = e^{-t} \qquad e_2 = f\left(1 - e^{-t}\right)$$

$$1 - \frac{N}{E} = e^{-t}$$

Solution der Stationarität

$$1 \ll P \ll N$$

$$\frac{de_2}{dt} \approx \frac{1}{1 + N e_1} - e_2 \qquad \boxed{e_2 = \frac{1}{1 + N e_1}}$$

$$0$$

$$\frac{1}{1 + P e_2} = e_1$$

$$\frac{1}{1 + \dfrac{P}{1 + N e_1}} = e_1$$

$$\frac{1 + N e_1}{1 + N e_1 + P} = e_1 \; \cancel{(1 + N e_1)}$$

$$1 + N e_1 = N e_1^2 + P e_1$$

$$\boxed{1 + (N - P) e_1 = N e_1^2}$$

Balans $\frac{N}{P} \gg 1$
$\quad N \gg 1$

$$e_1 = \frac{N - P}{N}$$

$$e_2 = \frac{1}{N - P}$$

$$\frac{1}{Ne_1} - e_2 = \frac{1}{Ne_1} e^{-t}$$

$$1 - (Ne_2)e_1 = e^{-t}$$

H

$$e_1 = 1 - \frac{1}{N} - \delta$$

$$\frac{1}{N} - e_2 e_1 = \frac{1}{N} e^{-t}$$

$$e_2 \quad\begin{cases} Ne_2 = 1 \\ \text{at } t = \end{cases}$$

$$\frac{1}{N} - e_2\left(1 - \frac{1}{N} - \delta\right) = \frac{1}{N} e^{-t}$$

$$1 - Ne_2\left(1 - \frac{1}{N} - \delta\right) = e^{-t}$$

if $Ne_2 = 1$

$$\boxed{\frac{1}{N} + \delta = e^{-t}}$$

$$e^{-\tau}$$

$$\delta = e^{-(t-\tau)}$$

$$\ln \frac{1}{\delta} = \Delta t$$

$$\ln \frac{1}{\delta}$$

$$\ln N = \tau_0$$

$$\ln \frac{\delta}{\delta}$$

$$-\log_{\delta} \frac{1}{N}$$

$$\boxed{e_1 = \frac{1}{2}}$$

$$1 - \frac{1}{2} = e^{-t_{1/2}}$$

$$1$$

$$e_2 = \frac{1}{N_{1/2}} - \frac{1}{N_{1/2}} e^{-t}$$

$$1 = 2 - 2e^{-t} = 2(1 - e^{-t})$$

$$\frac{1}{2} = 1 - e^{-t}$$

13 Amino acids are activated

$$APPP + \overset{NH_2}{R \cdot COO^-} \longleftrightarrow AP O C\overset{=O}{R} + PP$$

$$\downarrow NH_2OH \quad NH_3^+$$

$$AP + \overset{C=O}{\underset{R NH_2}{|}} \overset{}{_{NHOH}}$$

8000

12,000 [ATP + 5000]

two P bound 16,000

Uridilic + D.P. bal 5000

Novelli [autotroph enzymes out of bacteria
with couple adenine with AA
suggest believes to grow them with AA
limitation ———]

$$1 - (Ne_2)\, e_1 = e^{-t}$$

$$e_2 = \frac{1}{P}$$

$$e_1 = \varepsilon$$

$$1 - \frac{N}{P}\varepsilon = e^{-\tau}$$

$$\frac{N}{P}\varepsilon < 1$$

$$\boxed{\tau = \ln\left(\frac{1}{1 - \frac{N}{P}\varepsilon}\right)}$$

$$\varepsilon = \frac{1}{10}$$

$$\boxed{t \sim \frac{1}{\frac{N}{P}\varepsilon}} \quad \text{life time}$$

$$\frac{N}{P} = 2$$

13 Amino acids are activated

$$APPP + \overset{NH_2}{R \cdot COO^-} \longleftrightarrow AP \cdot O \cdot \overset{O}{C} R + PP$$

$$\downarrow NH_2OH \quad NH_3^+$$

$$AP + \overset{O}{C}\overset{}{-NHOH}$$
$$RNH_2$$

12,000 (ATP 5,000)

know P bond 16,000

unydol ↔ D.P. cal 5,000

Novelli [activate enzymes out of bacteria
with simple adenine with AA
suggest bacteria to grow them with AA
limitation —]

$$1 - (Ne_2)e_1 = e^{-t}$$

$$e_2 = \frac{1}{P}$$

$$e_1 = \varepsilon$$

$$1 - \frac{N}{P}\varepsilon = e^{-\tau}$$

$$\frac{N}{P}\varepsilon < 1$$

$$\boxed{\tau = \ln\left(\frac{1}{1 - \frac{N}{P}\varepsilon}\right)}$$

$$\varepsilon = \frac{1}{10}$$

$$\frac{N}{P} = 2$$

$$\boxed{t \sim \frac{N}{P}\varepsilon}^{-1} \quad \text{lifetime}$$

or 6.) RNA makes DNA

RNA

Sunday May 12/57

Duester bonds

3000 cal
single peptide bond
~ 2000 #

B₁ — c — c — c — s — c — op — o — P·O
          o   o

B₁ — c — c — c — c — c — op — o — P·O

DNA ↑↑
? RNA ↓↓ ?

I A_i A_j A_k A_0 $A_i - A_j - A_k - A - A -$ Hughes

A_i'' (...)

A_2''

DNA $AA - AA - AA$

$AA - AA$ $A - A - AA - AA$

$$N \quad O - M - O \qquad \overset{O}{\underset{O}{P}} - OH \quad - AA$$

$$B_1 \quad - C - C - C - C - CH_2$$
$$O \qquad O = P - OH$$

$$B_2 \quad - C - C - C - C - CH_2O$$
$$O \qquad O$$

$$B_3 \quad - C - C - C - C - CH_2O - P - OH$$

RNA

DNA

Demand at least 1 A or 1 U !

AUC
AUG
ACU
AGU
UAC
UAG
UCA
UGA
AUU
UAA

A, A +

Protein and

DNA makes ½ of RNA

RNA makes ½ of RNA

RNA

AA AA AA

C

Stanier

E
⟶ Tr

$R + Tr \xrightarrow{E} \{Tr-R\} \xrightarrow{E} \{Tr(\cdot)A-R\}$

Tr

5.) RNA makes $\frac{1}{2}$ RNA

$$E_i^{xx}(A_i, \text{ribose} -)$$

AL

6.) DNA makes $\frac{1}{2}$ DNA

$$E_{AA}^{*}(AA, \text{dexoxyribose} +)$$

4 RNA makes $\frac{1}{2}$ DNA

$$E_{AA}^{*}(AA, \text{dexoribose} -)$$

RNA

AA

RNA makes Protein A_iA_j Prot

2) RNA

$E_i^*(A_i; + ; ribose +)) E_j^*(A_j; -; ribose +)$

        mRNA

        (disorder—

1)   A   or $A_1 A_j - A_n$ Protein

RNA   RNA

or 3.) ½ proteins [RNA the proteins]

3.) RNA makes ½ of RNA

RNA

            AA   AA   AA-AA-AA   nucleoprotein(RN
                                          (ribose—

$E_{AA}^{**}(AA; ribose -)$ AA   DNA makes ½ RNA

4.)   AA   nucleoprot
DNA   (ribose +)

AA-AA   3 and 4   AA-AA-AA   polypeptide

                                    RNA

AA-AA   P.TO   AA-AA-AA   polypeptide

~~5.) RNA makes ½ of DNA~~

0.) A_i A_j A_k A_l $\longrightarrow$ A_i—A_j—A_k—A_l Protein

$\frac{1}{2}$ RNA

$E^*(A_i, \text{ribose} +)$ 20 enzymes

1.) a.) A_i A_j A_k A_l $\longrightarrow$ A_i—A_j—A_k—A_l Protein

2RNA $\longrightarrow$ RNA

$E^*(A_i, \text{ribose} +)$

RNA makes Protein

b.) RNA makes Protein turns over $\frac{1}{2}$ RNA

A_i A_j A_k A_l $\longrightarrow$ A_i—A_j—A_k—A_l Protein

RNA $\longrightarrow$ RNA destroyed

c.) A_i RNA makes $\frac{1}{2}$ RNA (independent)

A_i—A_k—A_k—A_l

$\frac{1}{2}$ RNA(+) $\longrightarrow \frac{1}{2}$ RNA(−) $\longrightarrow \frac{1}{2}$ DNA(+)

$\frac{1}{2}$ DNA(+), $\frac{1}{2}$ DNA(−) $\longrightarrow \frac{1}{2}$ RNA +

chang enzymatic levels hu

mutation

RNA(+) $\longrightarrow$ RNA(−)

preferable

RNA(−) $\longrightarrow$ DNA(+)

DNA(+) $\longrightarrow$ DNA(−)

DNA(−) $\longrightarrow$ RNA(+)

$\longrightarrow$ $\longleftarrow$

RNA(−) with DNA(+)
RNA(+)   & DNA(−)

$\longrightarrow$ $\longleftarrow$

2 Alternative:

$\frac{1}{2}$ RNA(−) $\longrightarrow \frac{1}{2}$ RNA(+) $\longrightarrow \frac{1}{2}$ DNA(−)

$\frac{1}{2}$ DNA(−) $\longrightarrow \frac{1}{2}$ DNA + $\longrightarrow \frac{1}{2}$ RNA(−) $\longrightarrow$ RNA + Prot.

Uninterrupted
has up RNA −

DNA makes $\frac{1}{2}$ RNA (nucleoprot)

E (AA, ribose ▬)

$\frac{1}{2}$ RNA(+) makes $\frac{1}{2}$ RNA (nucleoprot ▬)

E (AA ribose ▬)

RNA

RNA

nucleoprot
ribose (▬)

AA

AA — AA — AA

2

$\frac{1}{2}$ DNA(+) makes $\frac{1}{2}$ RNA(+)

*

E (AA, ribose +)

AA—AA—AA—AA

nucleoprot *
ribose +

3)

nucleoprot

AA AA AA AA

polypeptide

AA AA AA AA

RNA

nucleoprot

RNA

4)

DNA

AA
AA
AA
AA

AA
AA
AA
AA

AA
AA
AA
AA

AA
AA
AA
AA

polypep
tide

$\frac{1}{2} DNA(+)$ makes $\frac{1}{2}$ of $DNA(-)$

4,) $\quad \stackrel{+}{=} \{AA_2 \text{ desoxiribose } (-)\}$ H

$\frac{1}{2} DNA+$ $\frac{1}{2} DNA(-)$

nucleoprot desoxiribose —

5,) $\frac{1}{2} RNA(-)$ makes $\frac{1}{2}$ of $DNA(+)$

$\stackrel{+}{=} (AA \text{ desoxiribose } +)$

RNA RNA

nucleoprot desoxiribose (—)

6,) nucleoproteins $\qquad$ DNA

AA — polypeptide

Brak: Journal of Bacteriology 1956
tabumbda (methionin requiring mutant)

1)

comes off,   H

→ 🔴   🔴 does not come

⟍⟋   A B C ribos

RNA(-) A¹ B¹ C¹   A¹ B¹ C desoxy

DNA(+) ⟵   DNA(-) ⟵

E (A🔴ribose ╪ →)
        A B C ᴿ

E (AA → ribose ╪ ⟵)

E (AA(-) ribose — ⟵)

E (AA(+) desox ribose ╪ ←)

E (A🔴→ ribose +)

E (AA (←) ribose +)

E (AA ← ribose —)

E (AA ← desox ribose +)

E (AA ← desox ribose —)

Protein

RNA +

RNA −

AA - AA -

A G U C Rules .... (A G U*C) desoxiribose

A — U*
C   G — Ar
U* — A
C   G

→ Hur

— Ar

——— poly
A G G ⊕

A G U ⊖
(U C A)

Ai Aj — Ak
x x x   x x x   x x x   ½ RNA —   (½ RNA +
                                              →)
½ RNA (+) ½ RNA (-) → ½ DNA(+) → ½ DNA + ½ RNA (+)

in cytoplasm

these arrows
on DNA are
in opposite direction

$v = 10^4 \text{ cm/sec}$

$N\, 10^4\, 10^{-16}\, 10^{-3} = \int$

$N\, 10^{-15} = 1000$

$N = 10^{18} \text{ molecules}$

$\bar{t} = \dfrac{1}{t_{dip}} \dfrac{1}{t_o}$

$r = \dfrac{n_i}{\bar{t}}$

$r = 1000$

In terms of Avogadro

$10^{18} \text{ molecules/cc}$

$10^{21} \text{ molecules (liter)}$

for 100 gm (Hg) mol weight

$\dfrac{10^{23}}{6\, 10^{23}} = \dfrac{1}{6} \text{ gm/liter}$

Answer nucleotide concentration can be much higher than H H conc.

10^4 Molecules in 20 min $\boxed{1200}$ $\textcircled{4}$

enzymes

10 Molecules/sec

$\boxed{\text{100 Avogadro}}$

$\frac{1}{m}$ $\boxed{e^{-\frac{n}{2}}}$

$\sum_n \frac{x^k}{k!} = e^x$

$P(k) = \frac{n^k}{k!} e^{-n}$

$\sum_{k=0}^{\infty} P(k) = 1 =$

$\boxed{m e^{-rt}}$ $\boxed{1 - m e^{-rt}}$

$rt = n$

$W(t) = \frac{d}{dt}\left[1 - m e^{-rt}\right] = r m e^{-rt}$

$\bar{t} = \int_0^{\infty} r m t \, e^{-rt} \, dt = r m \frac{1}{r}$

$\int_0^{\infty} t e^{-rt} dt = -\frac{d}{dr} \int_0^{\infty} e^{-rt} dt$

$= -\frac{d}{dr}\left[\frac{1}{r}\left(e^{-rt} \frac{1}{r^2}\right)\right] =$

$\boxed{\bar{t} = \frac{m}{r}}$

$\frac{1}{2}$ RNA(-) makes $\frac{1}{2}$ DNA +

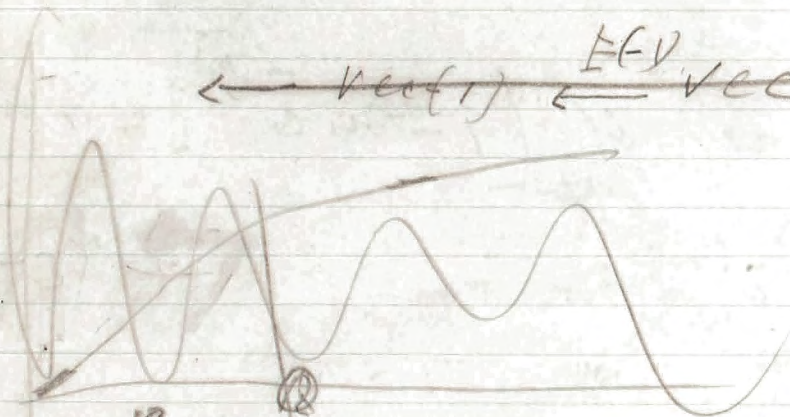$\frac{1}{2}$ DNA(+) makes and links up with $\frac{1}{2}$ DNA -

~~DNA~~ = DNA

DNA makes $\frac{1}{2}$ RNA(-)

Demand complementary code must
not be identical with code

$\boxed{A\,B\,a\,b}$ $1\;2\;3\;4 = 24$ give $\underline{\underline{12}}$

$M(2)\,E(2) \rightleftharpoons E(-1) \rightleftharpoons E$

$\xleftarrow{\hspace{1cm}} \;\;vec(1) \;\rightleftharpoons\; \overset{EE\,V}{vee} -$

invariant

$10^{13} e^{-\frac{Q}{RT}} =$

RNA+ →

RNA(−)
_____

←

RNA(−) ⟶ DNA(+), →

←   →

⟶ DNA(+) ══ DNA(−)   DNA (+) ══

RNA(−)-AA   RNA(−)-AA

←   RNA(+)

AA AAAA

[− − −] (−)   DNA makes RNA(−)-AA

[+ + +]

RNA-AA / makes RNA(+) naked   (+)
              + Protein   (+) (−)

In the nucleus (in bacteria)

RNA(−) (−)   RNA(−) (−)

AA-DNA(+) ══ DNA(−)-AA   DNA(+) ══

in the nucleus   DNA+

In the cytoplasm

T dependence of Z production in consolidation :—

N molecules
t_1, time it takes he assemble
t_2 " " " for evaporate if free
(1-f), fraction of those formed fixed & covered

$$\frac{1}{rate} = t_i + \frac{t_2}{const\ large}$$

$$rate = \frac{t\,const\,f\,t_1}{const\,f\,t_1 + t_2}$$

dimensions
$$[const] = \left[\frac{1}{t}\right]$$

$$const\,f\,t \gg 1$$

$$\frac{t}{t} = \frac{A}{const\,t}\,\frac{1}{t}\,e$$

~~~~~~~~~~~~

$$\overline{1000\ places = m}$$

$$m = \frac{1}{5}\,e^{\frac{t}{5}}$$ average machine time for one place

$$e^{-\frac{t}{5}} = e^{-t}$$

probable that $m-1$ are filled
probable that 1 is empty not filled

$$m\,e^{-t}\,(1 - m\,e^{-t})^{m-1}$$

$$\int t\,m\,e^{-t}\,\blacksquare\,(m-1)\,m\,e^{-2t}\,dt = \frac{1}{2}\int t\,e^{-2t}\,dt$$

$$m\,e^{-t} + 2(m-1)\,e^{-2t}\qquad m-$$

$$\int \frac{2(m-1)}{m}\,e^{-t}\,t\,e^{-t}\,dt$$

$\frac{1}{2} 10^{-12}$ gm Protein

Number of enzymes at 100000 a piece !

$\frac{1}{3} 10^{-12} \times 10^{-5} \times 6 \cdot 10^{23} = 2 \cdot 10^{6}$ enzymes molecules

or 10^{4} genes – 200 Molecules of enzyme

say $\underline{100}$

of 1000 genes have 1000 enzymes that leaves nothing or say 10 for the other 9000 genes

[1 gene <u>could</u> produce 10^{5} Molecules of enzyme per 30 min $\simeq$ 2000 sec $\sim 5 \cdot 10^{2}$ molecules/ $\frac{only}{sec}$]

needing say $5 \cdot 10^{3}$ molecules of arginine / sec

In general however assuming 100 molecules of enzyme 5 Arginine per sec needed
or at most 50

In general assuming at most 1000
enzyme level   5 Molecules of enz/sec

Assume lamps on $\frac{1}{10}$ sec
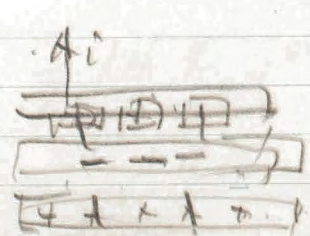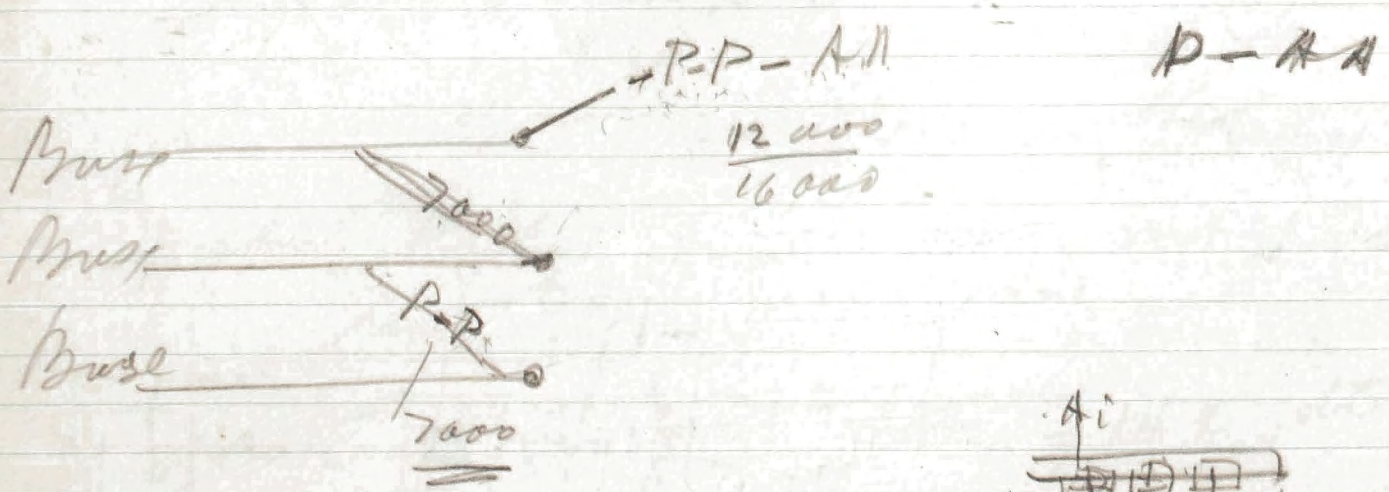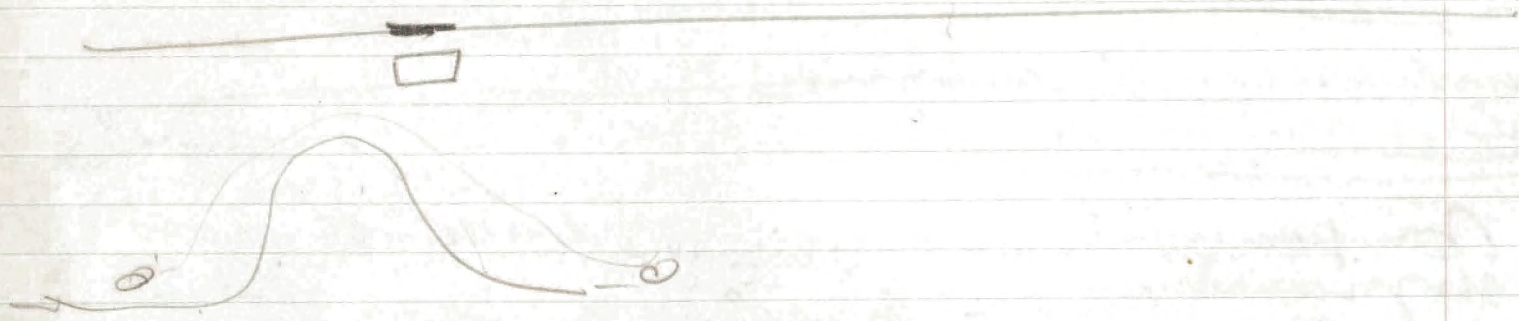
$$10^{13} \cdot e^{-\frac{Q}{RT}} = 10$$

$$e^{-\frac{Q}{RT}} = 10^{-12}$$

$$\begin{array}{c} 23 \\ 46 \\ \hline 27.6 \times 600 \end{array} \cong 12$$

$$\begin{array}{c} 48 \\ \hline 17,000 \end{array}$$

$$\boxed{7+7+9}$$

12 000 / 16 000

$P-P-AM$ $P-AM$

Brass

Brass   7000

Brass   P-P   7000

Ai

$$(1-e^{-t})^{m} = e^{-\frac{t}{m}} \quad \cancel{/} \; 11 = 1 - \frac{1}{m} \quad \textcircled{H}$$

$$1 - e^{-\frac{1}{m}} = \cancel{e^{-t}}$$

$$if \; \frac{1}{m} = e^{-t} = 1 - t$$

$$\boxed{t = \log m}$$

O.K.

per ampul mol weight $\underline{100}$

average time if they wait — if they want

10^{-5} mol/cc or $\cancel{6 \times 10^{18}}$ 10^{-8}/cc or $610^{-15} = N/cc$

$610^{-15} \; \frac{\cancel{10^{-15}}}{10^{-16}} (10^{-3}) \; 1510^{4} \, cm/se = 100$ or $\frac{1}{100} \underline{\underline{\;\;}}$

$\frac{\cancel{0.1} \, m}{10} \, sec \sim \frac{1}{10} \, sec$ $\quad 30 \, min = 1800.$ $\boxed{\cancel{1}\frac{1}{10} \; second \; hotter}$

Problem if each hangs on for Δt. work that

if $\cancel{\;\;}$ $m \, e^{-\frac{t}{\Delta t}}$ probab of 1 missing

probab of

Hangs on, assume $\frac{1}{10} \, sec$

$\cancel{\frac{1}{10}} \, e^{-\frac{Q}{RT}} = \frac{10^{\circ}}{10^{-13} \, 10^{12}}$ $\cancel{\text{MMMM}}$

$2 \times 7 + 16$ $\boxed{\frac{23}{\cancel{30,000}}}$

$DNA^+ \rightarrow RNA(-) \rightarrow DNA^+$ ⎰

DNA^- makes RNA^+ makes DNA ?

$30 \times 60 \times 10 = 18,000$

$$\boxed{AA = 1000}$$

Purine C

C5 N5
140

Pyrimidine
4C + 3N

Proline 5C 50

$\dfrac{1000}{10}$ [2 min

Proline

Base

Base

$BN - C - C - C - C - C - O - P - O - C - C - R$

OH

O

$- CO - AA$

$- O - P - O$

$\dfrac{1000}{100} = 10 \, \text{mol/min}$

Table

$$\require{cancel}\cancel{\dot{a}} - a2Ak\cancel{b}\,e^{-2Akt} = A\rho - 2Ak\,a\,e^{-2Akt} - 2Akb$$

$$b =$$

$$\text{if: } A\rho(m - f(t))$$

$$\text{rat } 2Akf$$

$$\dot{f} = A\rho m - A\rho f - 2Akf$$

$$= A\rho m - Af(\rho - 2k)$$

$$f = a\,e^{-\lambda t} + b \qquad a = -b$$

$$\dot{f} = -\lambda a\,e^{-\lambda t} = A\rho m - A(\rho - 2k)a\,e^{-\lambda t} - A(\rho - 2k)b$$

$$\lambda = A(\rho - 2k)$$

$$b = \frac{A\rho m}{A(\rho - 2k)}$$

$$f = b\left(1 - e^{-A(\rho - 2k)}\right) = \frac{\rho\, m}{\underset{\sigma}{\rho - 2k}}\left(1 - e^{-A(\rho - 2k)t}\right)$$

$$m - 1 \cancel{\tfrac{m \cdot Vf}{}} \frac{\rho\, m}{\rho - 2k}\left[1 - e^{-A(\rho - 2k)t}\right] \qquad \frac{2k}{\rho} \to 1$$

$$\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad k \cdot \frac{\rho}{2}$$

$$\boxed{1 - \frac{1}{m} = \frac{1}{1 - \frac{2K}{\rho}}\left[1 - e^{-A\rho\left(1 - \frac{2k}{\rho}\right)T}\right]}$$

$$1 - \frac{1}{m} - \frac{1}{1 - \frac{2k}{\rho}} = -\frac{1}{1 - \frac{2k}{\rho}}e^{-A\rho\left(1 - \frac{2K}{\rho}\right)T}$$

$$\left(1 - \frac{2K}{\rho}\right)\left(1 - \frac{1}{m}\right) - 1 = -e^{-A\rho\left(1 - \frac{2K}{\rho}\right)T}$$

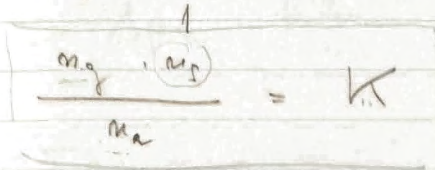$$\frac{1}{2} m \left(\overline{v_x^2} + \overline{v_y^2} + \overline{v_z^2} \right) = \frac{3}{2} kT$$

$$\frac{3}{2} m \overline{v_x^2} = \frac{3}{2} kT$$

$$\sqrt{\overline{v_x^2}} = \sqrt{\frac{kT}{m}}$$

$$e^* = \frac{1}{2} v$$

$$q + s \rightleftharpoons a$$

$$\underline{\frac{n_q \cdot \boxed{n_s}}{n_a} = K}$$

$$n_a = \frac{1}{2} n_q$$

$$e$$

$$\frac{2 n_q}{n_q} = K$$

$$\frac{1}{k_1} c \; e^* \rightleftharpoons [ec] \left(\frac{k_2}{k_1} \right) = k$$
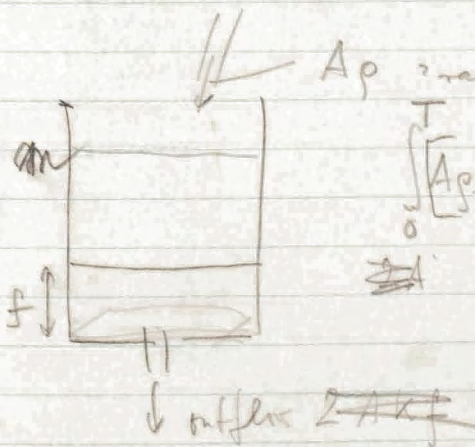
$$e^* + [ec] = e$$

$$(ec) = e - e^*$$

$$c e^* = (e - e^*) K$$

$$e^* (c + K) = e$$

$$k = \frac{c}{e}$$

$$e^* = \frac{e}{c + K}$$

$A \rho = $ rate of flow

$$\int_0^T [A\rho - 2Akf(t)] dt = $$

$\downarrow$ outflow $2Akf$

$$[A\rho - 2Akf(t)] dt$$

$$f(t + dt) = f(t) + [A\rho - 2Akf(t)] dt$$

$$f(t) + f'(t) dt$$

$$\frac{df(t)}{dt} = A\rho - 2Akf$$

$$f = a e^{-2Akt} + b = a e^{-2Akt} - 1$$

$$0 = a + b \qquad b = -a$$

$$f = a \left(e^{-2Akt} - 1 \right)$$

$$\frac{df}{dt} = -2Akt \, a \, e^{-2Akt} = -2Akt (f - 1)$$

$$\frac{2k}{\rho} = \mu \qquad\qquad \Delta T = \frac{\delta}{A\rho}$$

$$1 - \frac{1}{m} = (1 + \mu)\left(1 - e^{-A\rho(1-\mu)\left[\frac{1}{A\rho}\ln m + \frac{\delta}{A\rho}\right]}\right)$$

$$= (1 + \mu)\left(1 - e^{-(1-\mu)\ln m - \delta}\right)$$

$\varepsilon\left(1-\frac{1}{m}\right)-1=-e^{-Ap\varepsilon T}$ $\qquad$ $1-\varepsilon\left(1-\frac{1}{m}\right)>1$

$\ln\left[1-\varepsilon\left(1-\frac{1}{m}\right)\right]=Ap\varepsilon T$ $\qquad$ $\varepsilon\left(1-\frac{1}{m}\right)<1$

$T=\dfrac{1}{Ap\varepsilon}\ln\left[1-\varepsilon\left(1-\frac{1}{m}\right)\right]$ $\qquad$ $\left(1-\dfrac{2k}{\rho}\right)\left(1-\frac{1}{m}\right)<1$

$\qquad\qquad\qquad\qquad 0<\left(1-\dfrac{2k}{\rho}\right)<1$

$\dfrac{\partial T}{\partial k}=\dfrac{\partial T}{\partial \varepsilon}\cdot\dfrac{\partial \varepsilon}{\partial k}$ $\qquad$ $\dfrac{2k}{\rho}<1$

$\qquad\qquad\qquad\qquad \dfrac{\partial \varepsilon}{\partial k}=-\dfrac{2}{\rho}$

$-\dfrac{2}{\rho}\left[-\dfrac{1}{Ap\varepsilon^2}\ln[\ ]+\right.$

$\left.\qquad\qquad +\dfrac{1}{Ap\varepsilon}\dfrac{1}{1-\varepsilon\left(1-\frac{1}{m}\right)}\cdot-\left(1-\frac{1}{m}\right)\right]\rightarrow$

$\qquad -\dfrac{1}{\varepsilon}\ln(1-\varepsilon)+\dfrac{1}{1-\varepsilon}=0$

George W. Wheland — Resonance in Organic Chemistry II ed 1955 Appendix

martin)
Goutermann
Platt
(12 June)

1.2 benz
1.40 A
3A°

N
C

1.54 A

H H H H H H H H H H

Enzyme Antigen of Virus

F. Macfarlane Burnet

Cambridge Univ Press 1956

Burnham (Ken) Biochemical Journal (Engl.)
lost to Jems

Borah June 1956

Journal of Bact.

$\underline{\underline{Exp}}$.

Trypin/e

10^{-6} gm/cc $\quad 6 \times 10^{23} \times 0. \qquad 1.5 \ 10^{4} \ 10^{-15} \ 10^{-3} = 100$

$10^{24} \times 10$

$\frac{1}{100} \quad \times \log 1000 = \frac{1}{10}$

$\times t$

$e = 40$

$1.5 \times 23 = \underline{3} \qquad \qquad \times t = 3.5$

$\times = \frac{3.5}{40} \ 20 \ \frac{k}{6}$

$\frac{1}{5} \quad \frac{5}{1.44} =$

$\underline{10} \quad \textcircled{5}$

$\boxed{\text{J. B. C.}} \ \text{Kornberg (1956)}$

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